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**Background** 

**UGANDA ADVANCED LEVEL** 

# COMMON APPLICATIONS

**510: PHYSICS TEACHING SYLLABUS** 

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#### **UNITS:**

- Unit 1: <u>Fundamentals</u>
- Unit 2: <u>Dynamics I</u>
- Unit 3: <u>Properties of Matter</u>
  - Unit 4: <u>Dynamics II</u>



<u>Timings</u>

# Examples

Mitosis Digestive System Matrices Simple Cell Lightening Conductor Discharging Tube Human Eye Telecollaboration

# <u>Links</u>

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- Unit 5: <u>Geometric Optics</u>
- Unit 6: <u>Waves</u>
- Unit 7: <u>Thermal Properties of Matter</u>
- Unit 8: <u>Electrostatics, Electric Current and Electronic</u>

#### <u>Devices</u>

Unit 9: <u>Electromagnetism</u>

# **UNIT 1: FUNDAMENTALS**

Topics And Assessment Objectives:

- 1.1 Physical Quantities and their Units (9 Periods)
- Dimensions of fundamental quantities
- Fundamental units of ; mass (kg) , length(m), time (s), current (A), temperature(k), amount of substance (mol)
- Dimensions of derived quantities
- Use of dimensions to check equations
- Scalars and Vectors

25/10/2011 Columns & Rows Charts

### Examples

Refraction of Light Solving Polynomial Equations Workers' Database Base Converter

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- addition and subtraction of vectors by the component method
- resolution of vectors in two perpendicular directions
- components of a vector

## **Assessment Objectives:**

By the end of the topic, the student should be able to :

- State the basic physical quantities and state their SI units.
- Measure and read basic and derived physical quantities using linear and circular scales.
- Work out the dimensions of the derived physical quantities.
- Use Dimensions to check equations.
- Define scalars and vectors.
- Distinguish between scalars and vectors.

# 25/10/2011 Periodic Table

# <u>Links</u>

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# Examples

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• Solve problems involving vectors by the component method.



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# Examples

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# **UNIT 7: THERMAL PROPERTIES OF MATTER**

# COMMON APPLICATIONS

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- Temperature scales.
   Practical thermometers.
  - liquid in glass
  - constant volume gas thermometer

(9 Periods)

- electrical resistance
- thermocouples
- pyrometers
- Absolute temperature scale.

### **Assessment objectives**

Temperature

By the end of this topic, the student should be able to:

# Examples

Mitosis Diccotive D:/cd3wddvd/NoExe/.../meister10.htm List the different types of thermometers available for

moscuromont

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# Examples

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- of temperature and the thermometric properties used.
- Describe the steps involved in setting up a celcius scale of temperature.
- Describe the structure and action of the liquid- inglass, constant

volume gas thermometer, platinum resistance and thermocouple thermometers.

- Perform and describe measurement of temperature by the thermistor.
- Perform and describe measurement of temperature using a resistance of

an insulated copper coil and metre bridge.

- Perform and describe measurement of temperature using thermocouple.
- Compare temperature measured thermometers using different thermometric

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# Databases

**Tutorial** Introduction

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properties.

State the relative merits (advantages and disadvantages) of different

thermometers.

- Define the absolute temperature scale.
- Convert temperatures in degrees celcius to absolute temperatures.
- Describe and explain the structure and mode of operation of the optical

and total radiation pyrometers.

• 7.2 Specific Heat Capacity (9 Periods)

### Definition and its measurement.

- method of mixtures; Newton's law of cooling; cooling corrections
- electrical methods including; the continuous flow method for liquids.

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### **Assessment objectives**

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Quit

# Examples

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<u>Practical</u> <u>Questions</u>

# By the end of this topic, the student should be able to:

- Define specific heat capacity.
- Perform and describe an experiment to determine the specific heat capacity

## of a solid and a liquid by the method of mixtures.

- Explain what cooling correction is.
- Obtain a cooling correction in the method of

mixtures for the determination

of the specific heat capacity of a poor conductor of heat like rubber using the graphical method.

 Perform and describe an experiment to determine the specific heat capacities

of solids and liquids by electrical methods including the continuous flow method.

# HTML Editors Tutorial Basic html

<u>Netscape</u> Composer

# Examples

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# Art Principles of Art Elements of Art Coloured Pencil Paintings Pastel Paintings

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 State the advantages and disadvantages of the method of mixtures and the

continuous flow method.

- Solve problems involving heat loss and gain.
- Perfom and describe an experiment for comparison of specific heat

capacities of liquids using Newton's law of cooling.

• 7.3 Change of state.

- (9 Periods)
- Molecular theory explanation of melting, evaporation and boiling.
- Specific latent heat of fusion and vaporization.
   Internal and external latent heat of vaporisation.
- Electrical method of measuring specific latent heat of vaporization.

# Assessment objectives

By the end of this topic, the student should be able to:

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- Explain melting, evaporation and boiling using molecular theory.
- Define specific latents of fusion and vaporization.
- Perform and describe experiments for determination

of specific

latent heats using method of mixtures.

 Perform and describe an electrical method for determination of

specific latent heats of fusion and vaporization.

- Solve problems involving changes of state and changes in temperature.
- 7.4 Expansion of gases. (5 Periods)

- The gas laws and the equation PV = Nk<sub>B</sub>T or nRT
- Pressure and volume coefficients of expansion.

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#### By the end of this topic, the student should be able to:

- State Boyle's, Charles's and the pressure laws.
- Perform and describe experiments to verify the laws.
- Derive and use the equation PV = Nk<sub>B</sub>T or PV = nRT
- Define pressure and volume coefficients of expansion and show that

they are equal.

- 7.5 Kinetic Theory of gases. (10 Periods)
  - Brownian motion and evidence of molecules
  - Postulates about the molecules of an ideal gas.
  - Derivation of P =  $\underline{1}$  r <c<sup>2</sup>> and comparison with PV = Nk<sub>B</sub>T or nRT.

#### 3

- Deduction from the ideal gas equation.
  - Avogadros hypothesis

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- Graham's law of diffusion.
- Dalton's law of partial pressures.
- ♦ Real gases.
  - Van der Waal's equation of state.

```
( P + <u>a</u> )(V - b) = RT
v<sup>2</sup>
```

P vs V curves for a real gas.

critical temperature.

 Saturated and unsaturated vapours, saturated vapour pressure.

connection with boiling of a liquid.



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# **Background**

# **KINETIC THEORY (CON'T)**

# COMMON APPLICATIONS

### **Assessment objectives**

By the end of this topic, the student should be able to:

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• Explain what is meant by kinetic theory of gases.

 Explain quantitatively why a gas exerts pressure on the walls

of its container.

Derive the expression P = <u>1</u> r <c<sup>2</sup>> stating any assumptions made.

Animations & <u>Timings</u>

### Examples

<u>Mitosis</u> Digestive

System

Matrices

Simple Cell

Lightening

Conductor

**Discharging** 

<u>Tube</u> <u>Human Eye</u> <u>Telecollaboration</u>

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### 3

- Relate the mean kinetic energy of a gas to its absolute temperature.
- Use the formula P = <u>1</u>r <c<sup>2</sup>> to deduce Avogadro's Hypothesis,

### 3

Boyle's law, Charles's law, Dalton's law of partial pressures and Graham's law of diffusion.

- Distinguish between a real and an ideal gas.
- Account for the difference between equations PV =
   RT and

( P + <u>a</u> )(V - b) = RT v<sup>2</sup>

- Define critical temperature T<sub>c</sub> of a gas.
- Draw labelled P-V diagrams to show the behaviour of a real gas under

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# Examples

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<u>Light</u>

<u>Solving</u>

**Polynomial** 

<u>Equations</u> <u>Workers'</u> <u>Database</u> <u>Base Converter</u>

**Activities** 

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compression for temperatures above and below the critical temperature.

- Distinguish between a gas and a vapour.
- Distinguish between saturated an unsaturated vapour and define saturated

vapour pressure.(s.v.p)

- Explain the occurrence of saturated vapour pressure using molecular theory.
- Use kinetic theory to explain the effect of volume and temperature change

on s.v.p.

- Distinguish the behaviour of saturated vapours from that of unsaturated ones.
- Use Dalton's law of partial pressures to solve problems on s.v.p
- Relate variation of s.v.p to boiling point.
- Describe an experiment to measure the variation of s.v.p of water with

temperature.



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# Example

Periodic Table

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# Examples

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- 7.6 Thermodynamics
  - Work done by an expanding gas
  - Internal Energy
  - First law of thermodynamics DQ = DU + DW
  - Principle specific heat capacities, the relation Cp-Cv = R
  - Isothermal and adiabatic changes of a gas inlcuding work done
    - by a gas on such a process.

#### **Assessment objectives**

By the end of this topic, the student should be able to:

 State the component of the internal energy of a real gas and the factors

(12 Periods)

on which they depend.

 Define an ideal gas and show that the internal energy of an ideal gas has no 25/10/2011 Logos

> Practical Questions

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potential energy component.

 Explain the meanings of terms: isovolumetric, isobaric, isothermal, and

adiabatic changes.

 Derive the expression W = Pdv for the work done when a gas expands and

relate it to the area under the P-V curve.

- State the first law of thermodynamics and apply it to isobaric processes.
- Explain why a gas has more than one specific heat capacity.
- Define specific heat capacity of a gas at constant pressure and constant

volume.

Explain why the molar principle heat capacity at constant pressure Cp

is greater than that at at constant volume Cv



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- Derive the expression C<sub>p</sub>-C<sub>v</sub> = R
- Relate g = C<sub>p</sub>/C<sub>v</sub> to atomicity of a gas.
- Represent isovolumetric, isobaric, isothermal, and adiabatic processes on

### a P-V sketch.

 State the conditions necessary in practice to achieve isothermal and adiabatic

#### processes.

- State and use the equations relating
- Derive expressions for the work done in isothermal and adiabatic processes.
- Solve problems involving isovolumetric, isobaric, isothermal, and adiabatic

processes.



Thermal Conduction.

- Mechanism of thermal conduction in insulators and in metals.

- Thermal conductivity.
- The relation <u>DQ</u> = k A <u>DT</u>

Dt Dx

Measurement of thermal conductivity of good and bad conductors of heat.

Convection as a consequence of change of density.

- Radiation as a form of energy.
  - Blackbody radiation
  - Energy distribution in the spectrum of blackbody radiation.
  - Stefan's law  $E = sT^4$ .
  - Wein's displacement law, I<sub>m</sub>T = 2.9x10<sup>-3</sup> mK.
  - Surface temperature of the sun.
  - Survey of the electromagnetic spectrum.

#### Assessment objectives:

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#### By the end of this topic, the student should be able to:

 Explain the mechanism of heat conduction in gases, liquids metallic

and non-metallic solids.

 State the factors which determine the rate of heat transfer through

a material.

- Define thermal conductivity k of a material.
- Draw a sketch graph to show the variation of temperature with length

along a lagged and an unlagged metal bar.

Perform and describe an experiment to determine thermal conductivity

of a good conductor of heat like copper and a poor conductor of heat like glass.

- Colve problems involving conduction of heat

- Creating Learning Networks for African...
  - Describe and explain the process of convection.
  - State properties of infra-red radiation and describe how it can be detected.
  - Define a blackbody and blackbody radiation.
  - Describe how an approximate blackbody can be realised in practice.
  - Draw sketch graphs to show variation of relative intensity with wavelength

and describe their special features.

State and use Wein's displacement law and Stefan's law in calculations, including

the estimation of the temperature of the sun.

- State Prevost's theory of heat exchanges and apply it in calculations.
- Arrange the components of the electromagnetic spectrum in order of

decreasing wavelength.



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**Background** 

UNIT 8: ELECTROSTATICS, ELECTRIC CURRENT AND ELECTRONIC DEVICES

COMMON APPLICATIONS

Presentation • 8.1 Electrostatics.

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# • 8.1.1 Basic Electrostatics Phenomena. (8 Periods)

- Charging by friction.
- Types of charges.
  - use of electroscope to detect charge.
- Charge by induction.
- Distribution of charge outside and inside a hollow conductor

at constant potential.

- Principle of the Van der Graaf generator.
- Applications
  - lightning conductor, electrostatic screening,

paint spraying, and dust extraction.

### **Assessment objectives**

By the end of this topic, the student should be able to:

Distinguish between a conductor and a non conductor.

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- Perform an experiment to show that there are two types of charges.
- Explain gharging by electrostatic induction.
- Explain the attraction of an uncharged material by a charged body.
- Describe the structure and action of a Gold Leaf Electroscope.
- Explain how a gold leaf electroscope can be charged negatively or

positively.

 Describe how a gold leaf electroscope can be used to determine the

charge on a body

Describe Faraday's Ice Pail experiment and state the conclusions that

can be deduced from it.

Perform and describe an experiment to show the

.....



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Page Properties Printing Quit Creating Learning Networks for African... distribution of charge

on a charged conductor of different shapes.

 Explain corona discharge (action at points) and give an example of its

application.

 Describe the structure and operation of the Van de Graaf generator.

# 8.1.2 The Electric Field

- Electric fields and electric field lines.
- Force between point charges.
  - Coulomb's law.
- Electric Field Intensity.
  - electric field intensity of a point charge.
  - electric field intensity between charged parallel

metal plates.

- Electric Potential
  - relationship between electric potential and electric

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# Examples

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#### Creating Learning Networks for African... field intensity.

charges.

a point charge.

a number of point charges.

equipotential surfaces and electric field lines.

Calculate the force between two point charges.

Define electric field intensity, state its units and

patterns for different charge configuration.

Derive and use the relation between electric

potential and electric field intensity.

draw sketch diagrams to show the electric field

State the expression for the electric field intensity at

Calculate the electric field intensity at a point due to

Calculate force on a point charge due to a number of

electric potential at a point in the electric field of a point charge.

### Assessment objectives

By the end of this topic, the student should be able to:

State Coulomb's law of electrostatics.

Practical Questions

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Compare Coulomb's law with Newton's law of graviation.

8.1.3 Capacitors

- Capacitor and capacitance , the farad
- Polarization of dielectrics
  - the dielectric constant
- Parallel plate capacitor
- Factors which affect capacitance
- Series and parallel arrangement of capacitors
- Energy stored in a charged capacitor

### **Assessment Objectives**

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# By the end of this topic the student should be able to:

- Define the capacitance of a capacitor
- State the factors which determine capacitance of a capacitor

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- Explain the action of a dielectric using the molecular theory
- Explain what is meant by dielectric constant (relative permitivity) and dielectric strength.
- Perform and describe experiments to investigate the dependence of capacitance of a parallel plate capacitor on the area, A, of the plates,the separation,d, of the plates and

the nature of the dielectric material between the plates using a gold leaf electroscope.

- Perform and describe an experiment to measure dielecric constant of dielectric material.
- State and use the law of coservation of charge
- Derive and use expressions for effective capacitance of capacitors in series and in parallel.
- Derive and use the expression for energy stored in a charged capacitor.

### END OF S6 TERM 1

#### Electric Current

Electric current as flow of charge I = Q
 t

the ampere, the Coulomb, electric potential difference, the volt.

Electric power

- resistance and Ohm's law
- electric energy, kWh
- Power delivered to an ohmic circuit element
- Interule conversion of electrical energy with forms

Simple d.c circuits

- e.m.f of a source of electrical energy
- internal resistance
- conservation of charge at a junction in a circuit
- resistors in series and narallel

- Creating Learning Networks for African...
  - potential divider
  - mechanism of metallic conduction, current density i = nev
  - mechanism of the heating effect of an electric current
  - temperature coefficients of resistance
  - electrical resistivity, p: the relation R = pl

Α

Measurement of resistance and voltage

the Wheat-stone bridge and its applications including

measurement of temperature coefficient of resistance

 the potentiometer and its applications including measurement of voltage, current, thermocouple, e.m.fs, comparison of resistances.

Assessment Objectives:

Creating Learning Networks for African... By the end of this topic the student should be able to:

- Define an electric current and state its unit
- State the charge carriers in different types of conductors (metals, ionized gases, electrolytes, semi-conductors)
- Explain the mechanism of electric conduction in metals.
- Derive and use the relation between current and the drift velocity of electrons in metals I = nAVde
- Explain the causes of electrical resistance in metals and

identify the factors which determine resistance of

а

metallic conductor.

- Define the term electical resistivity and state its unit
- Explain the effect of temperature on resistance.
- Define temperature coefficient of resistance and state its unit.
- State Ohm's law and give examples of ohmic and non-ohmic

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conductors, and draw sketch graphs to show their I-V characteristic

curves.

- Perform and describe an experiment to verify Ohm's law for metallic conductors.
- State and use the law of conservation of current at a junction.
- Derive and use expressions for effective resistance of resistor in series and in parallel.
- List sources of e.m.f
- Explain what is meant by e.m.f, E, and internal resistance, r, of a cell
- Explain how the e.m.f and iunternal resistance of a cell change with time

and use.

- Derive and use the expression P = I2R
- Convert energy in joules into kWh.
- Convert electrical energy to other forms of energy.
- Derive the condition for maximum power dissipation

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in the external

resistance and the expression for efficiency, h.

- Derive and use the condition for balance of Wheatstone bridge
- Perform and describe an experiment to compare resistances using simple metre bridge
- Perform and describe an experiment to determine the resistivity, p, and

temperature coefficient of a resistance of a wire using a metre bridge.

 Explain why the Wheatstone bridge network is not suitable for comparison of two very high or very low

resistances.

- Solve problems on simple bridges including calculations of end-corrections.
- Explain the principle of a slide wire Potentiometer
- Perform and describe an experiment to calibrate a slide wire potentiometer.
- Perform and describe experiments to determine the internal resistance. r. of a cell. the e.m.f. F. of

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thermocouple using the slide wire potentiometer.

- Perform and describe experiments to calibrate an ammeter and voltmeter using a calibrated slide wire potentiometer.
- State the advantages and disadvantages of the potentiometer over an ordinary voltmeter for measurement of voltage
- State theadvantage of using a potentiometer instead of a Wheatstone bridge to compart resistances.

#### • 8.3 Electronic Derives

- The Vacuum diode valve
  - thermionic emission
  - anode current anode voltage chracteristics
  - incremental resistance of a diode
  - half wave rectification.
  - full wave (bridge) rectification

### The vacuum triode
- anode current-anode voltage characteristics
- anode current -grid voltage characteristics
- anode slope resistance, mutual conductance and amplification factor
- amplification by a triode voltage gain, A = yRL

Ra

+ RL

The p-n junction

- I-V characteristic
- half wave rectification
- full wave rectifier using semi-conducting diodes

#### The transistor

transistor characteristics

#### **Assessment Objectives:**

By the end of this topic the student should be able to :

= Evaluin the mechanism of thermionic omission

25/10/2011

- Creating Learning Networks for African...
  - Describe the structure and operation of a vacuum diode
  - Draw a sketch graph of the anode current -anode voltage characteristics of a thermionic diode and

explain its special features.

- Perform an experiment to obtain the I-V characteristic of a p-n junction and explain forward bias and reverse bias
- Explain half-wave and full-wave rectificatin and how they can be achieved
- Draw sketch graphs of the anode current-anode voltage and mutual characteristics of a triode.
- Define the terms anode resistance, mutual conductance and amplification factor of a triode.
- Derive and use the expression A = yRL for the voltage gain

Ra + RI

- Describe the structure of n-p-n and p-n-p type transistorl
- Perform experiments to obtain I<sub>B</sub> V<sub>BE</sub>, I<sub>c</sub> V<sub>CE</sub>

Creating Learning Networks for African... and  $I_c - I_B$  characteristics of transistor.

#### 9.1 Magnetic Effects of an Electric Current

- Idea of a magnetic field as a field of force due to current
  - carrying conductors or permanent magnet
- Force on a current
  - carrying straight wire.
  - Fleming's left hand rule
  - definition of magnetic flux density and the tesla

 Magnetic field patterns due to an electric current in a straight wire, circular coil and long solenoid.

- Motion of a charge particle in a uniform magnetic field
- Hall effect
  - the Hall probe

Tarana an a current corning call in a uniform magnetic field

#### rorque on a current carrying coil in a uniform magnetic field.

- moving coil galvanometer
- conversion of moving coil galvanometer into an ammeter and voltmeter
- Magnetic force between current carrying conductors
  - definition of ampere
  - simple form of current balance

#### **Assessment objectives**

By the end of this topic the student should be able to:

- Define a magnetic field
- Perform experiments to obtain the magnetic field patterns for a bar magnet, a current - carrying

straightwire, a current - carrying circular coil, and a current - carrying solenoid.

Perform an experiment to determine the direction of

al for a state and the second state and the second

the torce on a straight current carrying conductor in a magnetic field.

■ State, explain and use the expressions B = m<sub>0</sub> <u>L</u>,

 $B = m_0 NI$ , and  $B = m_0 nI$ 

#### 2pa 2R

for the magnetic flux density at a perpendicular distance a from a straight current carrying wire, at the centre

of a circular coil of N turns each of radius R and at centre of a long solenoid of n turns per metre.

Derive and apply the expression for the magnetic force between two long parallel current - carrying conductor

 Derive and apply the expression for the magnetic force between two long parallel current - carrying

conductor

- Define the ampere
- Describe a simple form of a current balance.
- Recall and use the expression F = Bqv sin q for the force on a particle of charge q, moving in a uniform magnetic field of flux density B.

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- Describe quantitively the motion of a charge particle in a uniform magnetic field.
- Explain the Hall Effect
- Explain how a calibrated Hall Probe can be used to measure magnetic flux density.
- Derive and use the expression t = BANI sin q for the torque on a current carrying coil in a magnetic field.
- Describe how a moving coil galvanometer can be converted into an ammeter and into a voltmeter.
- Calculate the value of the resistor required to convert a moving coil galvanometer into an ammeter or voltmeter.
- Describe how a moving coil galvanometer is converted into a ballistic galvanometer.



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**Background** 

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Equations of uniformly accelerated linear motion.

(18 Periods)

- Distance, speed-time graphs for uniformly and nonuniformly accelerated linear motion.
- Interpretation of area under a speed-time graph.
- Meaning of the slope of the tangent at a point on the distance-time ,

speed-time graphs.

I.2 Kinematics

- Motion of a body falling freely near the surface of the Earth
  - Acceleration due to gravity (g)

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#### **Examples** Mitosis

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- Motion of a projectile:
  - An example of motion due to uniformly velocity in one direction and uniform acceleration in the perpendicular direction.
  - Time taken to reach maximum height, time of flight
     T, range of a projectile.
- Relative velocity
  - Examples.

#### **Assessment Objectives:**

### By the end of this topic, the student should be able to:

- Define displacement ,speed, velocity and acceleration.
- Draw sketchs and interpret various motion graphs.
- Determine the distance travelled and the acceleration from the velocity-time graph.
- Derive and use the following expressions:
- v = u + at,  $s = ut + \frac{1}{2} at^2$  and  $v^2 = u^2 + 2as$
- Perform and describe an experiment to determine g

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## **Activities**

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 Derive and use expressions for time taken to reach maximum height, time of flight,

maximum height and range for a projectile.

Solve problems involving relative velocity.

## UNIT 2 : DYNAMICS 1

- **2.1 Newton's Laws of Motion and Momentum** (9 Periods)
  - Newton's laws of motion.
    - inertia
    - resultant force F =ma
  - Linear momentum and its conservation
  - Impulse and relation to change momentum
  - Elastic and perfect inelastic collisions.

#### **Assessment Objectives:**

# Databases

By the end of this topic. the student should be able to:

## **Introduction**

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- State and use Newton's laws of motion.
- Define linear momentum.
- Verify that linear momentum is conserved in a collision.
- Verify and use the Principle of conservation of linear momentum in collisions.
- Distinguish between elastic , inelastic and perfectly inelastic collisions.

• 2.2 Solid Friction (8 Periods)

- Laws of friction.
- Coeffients of static and kinetic friction.
- Motion of a body on a rough inclined plane.
- Molecular theory explanation of solid friction

#### **Assessment Objectives:**

By the end of this topic, the student be able to:

Perform and describe experiments to measure the coefficient of static

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and of kinetic friction.

- State and explain the laws of solid friction in terms of molecular theory.
- Solve problem involving motion of a body on rough surfaces.
- 2.3 Work, Energy and Power (15 Periods)
  - Work as a product of force and distance in the direction of force.
  - Work-energy theorem
  - Force-distance graphs
  - Kinetic and gravitationa potential energy
  - Elastic potential energy
  - Conservative forces
  - Energy conservation and conversion
  - Dissipative forces
  - Power as rate of transfer of energy; P = Fv

#### **Assessment Objectives.**



Creating Learning Networks for African... By the end of this topic, the student should be able to:

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- Define work, energy and power.
- State and apply the Principle of energy conversion.
- Relate work to the force-distance graph.
- Calculate work done in a number of situations.
- Derive and use the expressions K.E = ½ mv<sup>2</sup>, P.E = mgh
- Distinguish between kinetic energy and gravitational potential energy.
- Derive and use the relationship between work done and change in energy.
- Perform experiments to determine efficiency of a simple system.
- State the Principle of Conservation of mechanical energy and illustrate it

with examples.

- Solve problems involving conservation of mechanical energy.
- Derive and use the expression P = Fv



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• 2.4 STATICS (Periods 15)

- Parallel forces
  - Resultant of parallel forces, turning effect of forces and moment of a force, couples.
- Coplanar forces

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<u>Timings</u>

## Examples

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- Equilibrium of forces.
- Triangle of forces to represent forces in equilibrium.
- Principle of moments.
- Centre of gravity.
- Fluids in static equilibrium
  - Density, relative density.
  - Pressure at a point in a fluid.
  - Archimede's Principle
  - Floatation

#### Assessment objectives:

#### By the end of this topic, the student should be able to:

- Define centre of gravity.
- Calculate the resultant of parallel forces.
- Define and use moment of a force, couple and torque.
- State and use the conditions for equilibrium for a system under the

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## <u>Links</u>

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Creating Learning Networks for African... action of coplanar forces.

- Solve problems related to three coplanar forces in equilibrium.
- Define and use density and relative density.
- Derive and use the expression for pressure at a point in a fluid.
- State and use Archimede's Principle.
- State the Law of floation and use it to solve problems related to floating bodies.
- Perform and describe experiments involving the Principle of moments.

End of S5 Term 1 (Estimated time : 8 Weeks)

## **UNIT 3 : PROPERTIES OF MATTER**

- 3.1 Fluid Flow (9 Periods)
  - Streamline and turbulent flow.
  - Torminal valasity



## <u>Links</u>

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- Terminăi velocity
- Bernoulli's equation.
- Viscosity in liquids and its determination.
  - Poiseulle's and Stoke's law methods.
- Viscosity in gases.

#### **Assessment Objectives.**

By the end of this topic, the student should be able to:

Explain the terms steady (lamina, streamline) and

turbulent flow as applied to the motion of a fluid.

- Explain the effects of viscosity of an object moving in a fluid
- Define the terms velocity gradient and coefficient of viscocity of a viscous fluid

and state their units.

State Stoke's Law and use it to define the expression

for terminal velocity of a



#### Examples

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sphere in a viscous fluid.

- Perform and describe an experiment to mesure viscocity of a viscous liquid.
- Derive and use Bernoulli's equation P + ½ rv<sup>2</sup> + rgh
   = a constant
- Explain the applications of Bernoulli's Principles in the filter pumps, atomisers and

erofoil.

- Explain the effects of temperature on viscocity of liquids and gases.
- 3.2 Deformation of Solids (9 Periods)
  - Classification of Solids on the basis of strength,

stiffness, ductility and toughness.

- Stress-strain curve for ductile and brittle materials.
- Flastic and plastic hobaviour

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- - Hooke's law
- Work done in extension and compression.
  - elastic potential energy.

#### **Assessment Objectives**

#### By the end of this topic, the student should be able to:

- Explain the term strength, stiffness, ductility, toughness and elasticity.
- Define stress, strain, and Young's modulus and state their units
- Perform and describe an experiment to verify Hooke's law using springs,

draw a sketch graph of the stress-strain and show the following features:

limit of proportionality, elastic limit, yield point, breaking point.

Explain the special features of the stress-strain graph for a ductile material

- Distinguish the elastic behaviours of ductile and brittle materials.
- Compare the elastic behaviours of ductile materials, rubber and brittle materials.
- Perform and describe an experiment to determine Young's modulus of a metal in

form of an elastic material.

- Relate the work done to the elasic potential energy.
- Relate the work done to area under the forceextension curve.

#### • 3.3 Surface Tension (12 Periods)

- Simple surface tension phenomena.
- Molecular theory of matter
  - Explanation of surface tension.
- Definition of surface tension.
- A Drassura difference across a subarical surface

- Creating Learning Networks for African...
  - Angle of contact
  - Capillary rise.
  - Methods of measuring surface tension.
  - Effects of temperature on surface tension.

#### **Assessement Objectives**

By the end of this topic, the students should be able to:

- Perform and describe experiments to show the existance of surface tension.
- Explain surface tension in terms of molecular theory of matter.
- Define surface tension and state its units.
- Derive and use expressions for excess pressure inside air and soap bubbles.
- Describe an experiment to measure angle of contact q.
- Explain capillarity.
- Dariva and use the expression **b 3g age a**

Derivé and use the expression n = <u>zg cos</u> q

rgr

- Perform and describe an experiment to measure surface using capillary.
- Explain the effects of impurities on surface tension.



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**UNIT 4: DYNAMICS II** 

#### 25/10/2011 COMMON APPLICATIONS

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- 4.1 Circular Motion (12 Periods)
  - Speed of a body, v = rw, moving with uniform angular speed

in a circle of radius r.

• Centripetal acceleration,  $a = v^2 = rw^2$  and centripetal force.

Examples

- Conical pendulum
- Banking of a road
- Upsetting and skidding
- Translation and rotation kinetic energy.

#### **Assessment Objectives**

## Examples

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By the end of this topic, the student should be able to:

- Define centripetal force.
- Define the radian.
- Derive and use the expression v = rw.
- Derive and use the expression a = v<sup>2</sup>/r = rw<sup>2</sup> and state its direction.

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Refraction of Light

## <u>Solving</u> Polynomial

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- Use the expression F = mrw<sup>2</sup> = mv<sup>2</sup>/r for centripetal force.
- Explain the following as applied to circular motion:
  - conical pendulum
  - banking of a road
  - motion in a vertical circle.
- Describe conditions for skidding and toppling of a cyclist or a vehicle

moving round a bend.

- Define moment of inertia about a given axis.
- State the expression for rotational K.E of a body rotating about an axis with a

constant angular velocity.

- Distinguish between transitional and rotational K.E
- Relate work done by a couple to rotational K.E

• 4.2 Gravitation (18 Periods)

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- ♦ Kepler's law's.
- Newton's law's of gravitation.
- ♦ Gravitational field including local variations of g.
- Principle of laboratory determination of G.
- Gravitational Potential
- Satellites
  - Mechanical energy in a given orbit.
  - Parking satellites.

#### Databases Tutorial Introduction

### **Assessment Objectives**

By the end of this topic, the student should be able to:

*Example* Periodic Table

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- State Kepler's laws
- State Newton's law of Gravitation.
- Derive dimensions of the gravitation G.
- Derive and use the relation between G and g.
- Describe the principle of laboratory determination of G.
- Derive and use Kepler's third law T<sup>2</sup> a r <sup>3</sup>.
- Define and use gravitational potential.

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- Define the velocity of escape velocity
- Derive and use the expression v<sub>e</sub> = ÖR<sub>e</sub>g
- Describe the variation of g from the centre of the earth to a point

above the earth's surface.

 Derive and use the formulae for K.E, P.E and mechanical energy of

a satellite in orbit.

- Define parking orbit and relate it to communication satellites.
- Derive and use the expression T<sup>2</sup> = <u>4pR</u><sup>3</sup> for parking orbit./

Gr<sup>2</sup>

- Explain a state of weightlessness.
- Define free fall.
- Perform and describe an experiment to determine the acceleration

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#### Examples

#### 4.3 Oscillations

Periods)

#### 4.3.1 Simple Harmonic Motion (SHM)

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of free fall.

- A special periodic motion defined by  $a = -w^2x$ .
- Derivation of the equation  $a = -w^2x$ .
  - a mass on a helical spring.
  - a simple pendulum.
  - a floating cylinder.
  - a liquid in a U-tube.
- Solution of  $a = -w^2 x$  of the form  $x = A \sin wt$  or x =Acos wt
- Graphical representations of displacement, speed and acceleration

in SHM.

Phase difference demostrated with two oscillating pendula or two masses

oscillating at the end of helical springs

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Amplitude, Period and frequency.

speed v =  $\pm$  wÖ(A<sup>2</sup> - x<sup>2</sup>)

- Interchange of kinetic and potential energy in SHM.
- Conservation of Energy.
- Measurement of acceleration due to gravity using
  - a simple pendulum.
  - a mass of a helical spring.



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## Simple Harmonic Motion (continued)

#### **Assessment objectives**

By the end of this topic, the student should be able to:

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Derive the expression a = -w<sup>2</sup>x component of acceleration of

a body moving in a circle.

- Define simple harmonic motion.
- Verify that a simple pendulum, a mass at the end of a string, a liquid

in a U-tube, floating cylinder and car piston oscillate with SHM.

- Define the terms period and amplitude.
- Derive and use the expression for the period in each of the above

25/10/2011 Digestive System

> <u>Matrices</u> <u>Simple Cell</u> <u>Lightening</u> <u>Conductor</u> <u>Discharging</u> <u>Tube</u> <u>Human Eye</u> <u>Telecollaboration</u>

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examples of SHM.

Verify that the solutions of the equation a = -w<sup>2</sup>x are of the form

#### x = A sin wt or x = A cos wt .

Explain phase difference between two different simple harmonic

oscillators.

 Draw sketch graphs to show the variation of displacement, velocity,

acceleration with time.

 Derive and use the expression v = ± wÖ(A<sup>2</sup> - x<sup>2</sup>) for the velocity of a simple

harmonic oscillator.

 Derive and use expressions for potential energy and kinetic energy of a simple 25/10/2011 <u>Refraction of</u> <u>Light</u> <u>Solving</u> <u>Polynomial</u> <u>Equations</u> <u>Workers'</u> <u>Database</u> <u>Base Converter</u>

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harmonic oscillator and hence the mechanical energy.

 Describe the interchange between kinetic energy and potential energy during

SHM and show that mechanical energy is constant.

 Draw a sketch graph to show the variation of P.E and K.E and mechanical energy

with displacement.

#### • 4.3.2 Damped and Free Oscillations.

- Damped oscillations.
- Forced oscillations and resonance.
  - practical examples.

*Example* Periodic Table

#### Assessment objectives

By the end of this topic, the student should be able to:



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- Distinguish between free and damped oscillations.
- Describe practical examples of damped oscillations with
- particular reference to the degree of damping and the importance
- of critical damping in such cases as car suspension system.
- Explain forced oscillations and describe practical examples of forced
- oscillations and resonance.
- Describe graphically how amplitude of forced oscillations varies with
- frequency.
- Define resonance.
- State the factors which determine the frequency response and sharpness
- of the resonance of a forced oscillator.
- List examples od cases where resonance is useful and where it is undesirable.
- Perform and describe an experiment to determine acceleration due to gravity using ;

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- (i) simple pendulum.
- (ii) helical spring.
- Perform an experiment to determine Young's modulus of wood using a

vibrating wooden beam .

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End of S5 Term 2 : Estimated time : 10 weeks (9 Periods per week)



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#### **Background**

**UNIT 5: GEOMETRIC OPTICS** 

## COMMON APPLICATIONS

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Digestive

<u>System</u>

<u>Matrices</u>

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## • 5.1 Reflection of light at plane surfaces. (6 Periods)

- Regular and diffuse reflection.
- Rotation of a plane mirror with direction of incident ray.
- Images formed in a plane mirror.
- Focal point, focal length, centre of curvature.

#### **Assessment objectives**

#### By the end of this topic, the student should be able to:

- Define a ray of light.
- Define a beam of light and draw sketch diagrams for the convergent and

#### divergent beams.

- Perform and describe an experiment to illustrate the principle of reversibility
  - of light.
- State the laws of reflection of light.

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 Perform and desctibe an experiment to illustrate the laws of reflection of

light.

- Distinguish between regular an diffuse refleciton.
- Perform and describe an experiment to determine the relation between angle

of rotation of a plane mirror and angle of the reflected ray while keeping the direction of the incident ray fixed.

- Describe the application of rotation of a plane mirror in the light beam galvanometer.
- Perform and describe an experiment to find the position and nature of an image formed

#### by a plane mirror.

 Derive and use the expression relating the number of images formed by two inclined

mirrors and the angle between the two mirrors.



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• List uses of plane mirrors.

5.2 Reflection of light at curved surfaces. (9 Periods)
The equation 1+1 = 1 = 2 u v f r
Spherical aberration, caustic surface.
Assessment objectives
By the end of this topic, the student should be able to:
Describe the types of curved mirrors.

> Define the terms focal point, centre of curvature, radius of curvature,

pole and aperure as applied to curved mirrors.

- Derive and use the relation r = 2f for curved mirrors.
- Distinguish between marginal and paraxial rays.
- Describe the formation of caustic surface.
- Describe spherical aberration and the use of

т i• •

Databases Tutorial Introduction

*Example* Periodic Table

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Creating Learning Networks for African... parabolic mirrors to

correct the defect.

Derive and use the formula 1 + 1 = 1 = 2

uvfr

 Determine image and object position, focal length and radius of curvature

by construction and calculation.

- Distinguish between the nature of images formed by convex and concave mirrors.
- Perform and describe experiments to detemine focal length or radius of curvature

of curved mirrors.

Perform and describe experiments to determine

focal length or radius of curvature

of curved mirrors using a distant object, no-parallax and an illuminated object.

List applications of concave and convex mirrors.

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#### • 5.3 Refraction of light at plane boundaries (9 Periods)

- Snell's law.
- Real and apparent depth.
- Critical angle and total internal reflection.
  - applications, including optical fibres.

#### **Assessment objectives**

By the end of this topic, the student should be able to:

Art

Principles of Art Elements of Art Coloured Pencil Paintings Pastel Paintings Book cover Posters Perform and describe experiments to demonstrate

refraction of light through

a glass block and through a liquid.

- State the laws of refraction.
- Perform and describe an experiment to establish
  <u>sin i</u> = constant

sin r

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- Define refractive index and explain its variation with optical media.
- Define refractive index in terms of velocities of light in the respective media.
- Derive and use relation gna = 1

a<sup>n</sup>g

 Derive and use the the expression 1n3 = 1n2 x 2n3 for three parallel sided

transparent media.

- Derive and use the expression n sini = constant.
- Perform and describe an experiment to determine the refractive index by the

apparent depth method.

- Explain critical angle and the total internal reflection.
- State the conditions for the occurence of total internal reflection.
- List applications of total internal reflection e.g fibre

Creating Learning Networks for African... optics, radio wave transmission,

binoculars, periscopes and mirage formation.



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#### **Background**

• 5.4 Refraction through Prisms (9 periods)

## COMMON

Minimum deviation

#### 25/10/2011 APPLICATIONS

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- Deviation by thin prisms.
- Prism spectrometer and white light spectrum.

#### **Assessment objectives**

## By the end of this topic, the student should be able to:

- Explain dispersion of white light by a prism.
- Derive and use the expression for deviation

 $d = (i_1 - r_1) + (i_2 - r_2)$ 

- Derive and use the expression d = (n 1)A for a small angled prism.
- State conditions required for minimum deviation to occur.
- Derive and use the expression n = sin(<u>A+D</u>min) / sin
  <u>A</u>

# 2

2

Perform and describe experiments to measure angle
 A of

25/10/2011 Discharging <u>Tube</u> <u>Human Eye</u> <u>Telecollaboration</u>

# <u>Links</u>

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a prism and  $\mathbf{D}_{\mbox{min}}$  using a spectrometer.

 Perform an experiment to determine refractive index of the material of the prism

using optical pins and spectrometer.

- Describe applications of glass prisms.
- 5.5 Refraction through a thin lens. (12 Periods)
  - Types of lenses.
  - Focal points, focal lengths
  - Power of lens.
  - ◆ Thin lens formula <u>1</u> + <u>1</u> = <u>1</u>
    - v f
  - Transverse magnification.
  - Displacement formula, L<sup>2</sup>-d<sup>2</sup> = 4d<sup>1</sup>
  - Conjugate foci; Newton's formula xy = <sup>2</sup>

...

• Full lens formula  $\underline{1} = (n-1) (\underline{1} + \underline{1})$  for a thin lens in air.

r1 r2

Methods of determination of focal lengths of both

25/10/2011 Database Base Converter

<u>Activities</u>

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converging and

diverging lenses.

- Defects of images formed by a lens;
  - spherical and chromatic aberration.

# <u>Links</u>

#### **Assessment objectives**

Databases Tutorial Introduction

*Example* Periodic Table

<u>Links</u>

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By the end of this topic, the student should be able to:

- Identify converging and diverging lenses.
- Define the terms:- principle focus, principle axis, optical centre,

focal length as applied to converging and diverging lenses.

- Draw ray diagrams to illustrate formation of real and virtual images.
- Derive and use the expression <u>1</u> + <u>1</u> = <u>1</u> = <u>2</u>
  - u v f r
- Define transverse magnification.
- Derive and use the relation: m = v

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# Examples

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Practical Questions

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#### u

- Derive and use the expression  $L^2 d^2 = 4d_1^4$
- Derive and use the expression <u>1</u> = <u>1</u> + <u>1</u> for thin lenses in contact.

# F ¦1 ¦2

- Define power of the lens and state its units.
- Derive and use Newton's formula xy = <sup>1</sup>/<sub>2</sub>
- Derive and use the expression  $\underline{1} = (n 1)(\underline{1} + \underline{1})$   $r_1$   $r_2$
- Explain chromatic aberration and spherical aberration in lenses and how

#### they are minimised.

 Perform an experiment to demostrate chromatic aberration and spherical

#### aberration in lenses.

Perform and describe experiments to determine

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Creating Learning Networks for African... focal length and radii of the

surfaces of the lenses.



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(9 Periods)

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## **Background**

# COMMON APPLICATIONS

- Magnifying power of an optical instrument.
- The magnifying glass (simple microscope).
- The compound telescone

5.6 Optical Instruments

## Presentation Software Tutorial Introduction Background

<u>Text</u> <u>Slides</u> <u>Graphics</u> <u>Animations &</u> <u>Timings</u> Creating Learning Networks for African...

- - magnifying power in normal adjustment.
- Astronomical telescope and Galiliean telescope.
  - magnifying power in normal adjustment.
- Reflecting telescope.
- Prism binoculars.
- Simple lens camera.
- The human eye.
  - eye defects and their corrections.

#### **Assessment objectives**

#### By the end of this topic, the student should be able to:

- Descirbe the optical parts of the human eye.
- Distinguish between long and short sightedness.
- Explain how the eye defects are corrected.
- Define the terms visual angle, angular magnification,

near point and far point.

 Describe structure and action of: simple microscope, compound

Examples

<u>Mitosis</u> Digestive

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# <u>Tube</u> <u>Human Eye</u> <u>Telecollaboration</u>

# <u>Links</u>

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# Examples

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microscope (normal adjustment) ,astronomical telescope, Galilean telescope,

reflecting telescope, simple lens camera, projector lantern.

State the advantages and disadvantages of reflecting telescopes

over refracting telescopes (Astronomical telescope and Galilean telescope)

- Derive and use the expression
  - M = ↓ for telescopes

- Define the eye-ring of a telescope and explain it's significance.
- Describe the action of prism binoculars.

# **UNIT 6: WAVES**

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**Base Converter** • 6.1 Types of Wave Motion. (9 Periods)

**Activities** 

<u>Links</u>

Databases Tutorial Introduction Transverse and longitudinal wave motion.

- Relation between v, ¦, and l.
- Progressive waves.
  - the equation for th eprogressive wave.

y = A sin 2p( <u>t</u> - <u>x</u> ) T l

- Transmission of energy by waves.
- Relation between intensity, frequency and amplitude.

*Example* Periodic Table

#### **Assessment objectives**

By the end of this topic, the student should be able to:

 Describe longitudinal and transverse waves and explain their mode

of propagation.

 Define the terms: displacement, amplitude, period, frequency and

<u>Links</u>

# Word Processing Tutorial

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# Examples

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Practical Questions

# **HTML Editors**

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wavelength.

- Derive and use the expression v = ||.
- Perform and describe experiments to demonstrate progressive wave.
- Explain phase of vibrations.
- Derive and use the expression y = A sin 2p(<u>t</u> <u>x</u>) and explain the

# ТΙ

significance of  $\pm$  in the equation.

Relate amplitude and frequency with energy.

# • 6.2 Superposition of waves

- Principle of superposition
- Stationary waves and their properties.
- Interference and beats, Doppler effect.
- Longitudeinal stationary waves and air columns, resonance.
- Stationary waves and stretched strings including the relation
  v = <u>1</u>Ö(T/m)

2L

Overtones and harmonies.

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<u>image</u>

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#### **Assessment objectives**

#### By the end of this topic, the student should be able to:

- State and use the principle of superposition of waves.
- Explain interference fo waves and occurrence of beats.
- Derive and use the expression for beat frequency.
- Explain Doppler effect.
- Derive and use the expression for Doppler shift.
- Explain the formation of a stationary wave.
- Explain the terms node and antinode.
- Perform and describe experiments to demonstrate formation of

stationary waves.

Derive and use the relation v = <u>1</u>Ö(T/m) for a stationary wave in

**2L** 

r . ..

a stretched string.

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Creating Learning Networks for African...

- Describe applications of stationary waves in strings.
- Perform and describe experiments to demonstrate longitudinal stationary

waves in air columns using open pipes and closed pipes.

- Demonstrate and explain resonance.
- Explain overtones and harmonics.
- Perform and describe experiments to measure velocity and frequency of

sound, using open and closed pipes.

 Derive and use a relationship between the frequency and length of an air

column.

 Perform and describe experiments to show variation in speed of sound in

different media and explain the variation.

Explain the dependence of speed of sound in air on

Creating Learning Networks for African... pressure, temperature

density and direction of wind.



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**Background** 

• 6.3 Wave theory of light

(18 Periods)

# COMMON APPLICATIONS

Presentation

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# Examples

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Creating Learning Networks for African...

- Huygen's construction as applied to reflection and refraction.
- Speed of light in air.
- ✤ Interference of light and its applications.
  - conditions for interference.
  - Young's double slit interference; derivation of

l = <u>aDx</u>

#### D

- thin films.
- Diffraction
  - demonstration of diffraction using water wave in a ripple tank.

single slit diffraction of light.

 plane transmission grating and application of formula dsinq = nl

in the determination of wavelength.

- Polarisation
  - as a phenomenon associated with transverse waves.

25/10/2011Creating Learning Networks for African...Ligntening<br/>Conductorproducing polarised light by reflection, doubleDischarging<br/>Tube<br/>Human Eye<br/>Telecollaborationrefraction,<br/>selective absorption and scattering.application of polarisation.

#### **Assessment objectives**

## <u>Links</u>

By the end of this topic, the student should be able to:

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<u>Worksheets</u> <u>Columns & Rows</u> <u>Charts</u>

# Examples

Refraction of Light Solving

<u>Polynomial</u>

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- Define a wave front.
- Distinguish between circular and plane wave fronts.
- Describe Huygen's construction.
- Apply Huygen's construction to reflection and refraction of

light

- Describe a terrestrial method of measuring the speed of light.
- Explain the terms coherent sources of light, path difference and

25/10/2011 <u>Equations</u> <u>Workers'</u> <u>Database</u> <u>Base Converter</u>

<u>Activities</u>

<u>Links</u>

Databases Tutorial Introduction

*Example* Periodic Table

<u>Links</u>

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Creating Learning Networks for African... optical as applied to light.

- Explain interference by "division of wave fronts".
- Describe Young's double slit experiments.
- Explain destructive and constructive interference.

Derive and use the equation I = aDx

#### D

Describe an experiment for measuring I using the double slit

arrangement.

- State factors which determine the appearence of fringes.
- Explain interference by "division of amplitude".
- Explain interference of light waves in thin films.
- Perform and describe an experiment to demostrate diffraction of water

waves in a ripple tank.

- Explain diffraction of waves.
- Describe plane transmission grating.

Introduction Text Page Properties Printing Quit

Examples

Address list Time table Newsletter Repeat Patterns Logos

Practical Questions

Tutorial Basic html Netscape Creating Learning Networks for African...

- Use the expression dsing = nl
- Describe an experiment to determine I using diffraction grating.
- Explain the terms polarisation, polarised light and polaroid.
- Describe polarisation of light by: reflection, double refraction, selective

absorption and scattering methods.

- State and use Brewster's Law.
- Explain applications of polarisation e.g. in sacce harimetry, photoelasticity,
- reducing light intensity and 3-D pictures (holography)



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#### Examples

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