Susceptibility of aluminium alloys to alcoholate corrosion in alcohol blended biofuels – technological influences

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Agenda



- Alcoholate corrosion of aluminium
- Technical temperature threshold
- Influences on initiation and propagation of alcoholate corrosion
 - Fuel composition
 - Temperature
 - Water content
 - Cyclic mechanical loading
- Conclusion



Corrosion mechanisms of metals in oxygenated fuels



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Electrochemical corrosion

due to <u>dissolved water</u> (e.g. low water content)

due to <u>separated water</u> (high water content) depending on material and fuel composition



pitting corrosion of X90CrMoV18 in an E10 gasoline blend



corrosion at the bottom of a tank after phase separation of water

Chemical corrosion "dry corrosion", chemical alcoholate corrosion

concerns alcohol containing fuels, in particular Al-, Mg-alloys and lead



"dry" alcoholate corrosion of a cast Al-alloy in an E10 gasoline blend

Alcoholate corrosion of aluminum



- Refers to the chemical corrosion of metals in the presence of fuel alcohol containing fuel blends
- Alcohols can react with aluminium alloys, lead and magnesium with the formation of alkoxide or alcoholate corrosion products

$$3C_{2}H_{5}OH + AI \rightarrow (C_{2}H_{5}O)_{3}AI + 3/2H_{2}$$
(1)
$$(C_{2}H_{5}O)_{3}AI + 3H_{2}O \rightarrow AI(OH)_{3} + 3C_{2}H_{5}OH$$
(2)
$$(C_{2}H_{5}O)_{3}AI \rightarrow AI_{2}O_{3} + 6C_{2}H_{4} + 3H_{2}O$$
(3)

(1) alcoxides (alcoholate) get hydrolyzed (2) or decomposed (3)

Damaging process may progress rapidly and is accompanied by an increase in pressure due to hydrogen formation

Alcoholate corrosion of Al



Aim of the investigation



Investigation of major influences on the alcoholate corrosion of aluminum alloys, like

- Material composition
- Fuel composition
 - Ethanol content
 - Water content
- Temperature
- Cyclic mechanical loading





- Defined as a guidance level (not a characteristic material value) for a specific alloy
- For a given level and type of alcohol of a fuel 🖁 blend
- Illustrates the effect of technical parameters and processes on the susceptibility to alcoholate corrosion

mixing sequence positive effect 41 negative effect

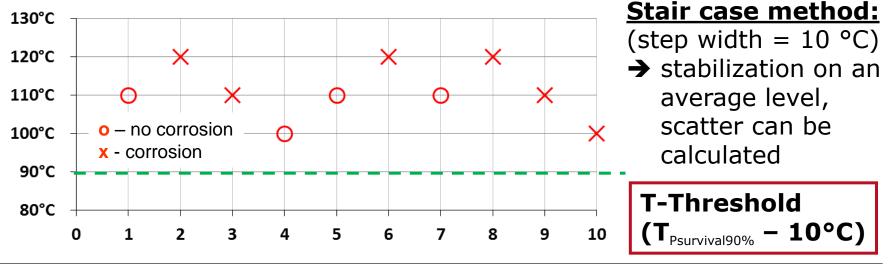
water

- defect free coatings (e.g. anodisation, NiP, solgel...)
- fuel + EtOH + water
- mixing of fuel in ambient air
- steric complex or long chained alcohols
- wear, scratching
- detrimental microstructural features
- aggressive impurities
- coating defects
- higher surface/volume ratio
- mixing sequence EtOH + water + fuel
- time
 - mechanical loads



Test conditions and deduction of the technical temperature threshold

- Preparation of fuels (e.g. reference fuel + SeccoSolv EtOH)
- Testing of partially immersed samples (168 h) in pressure tight vessels with protection against explosions
- Pressure signal is used as indicator for alcoholate corrosion
- Test sequence according to stair case method (n > 10)





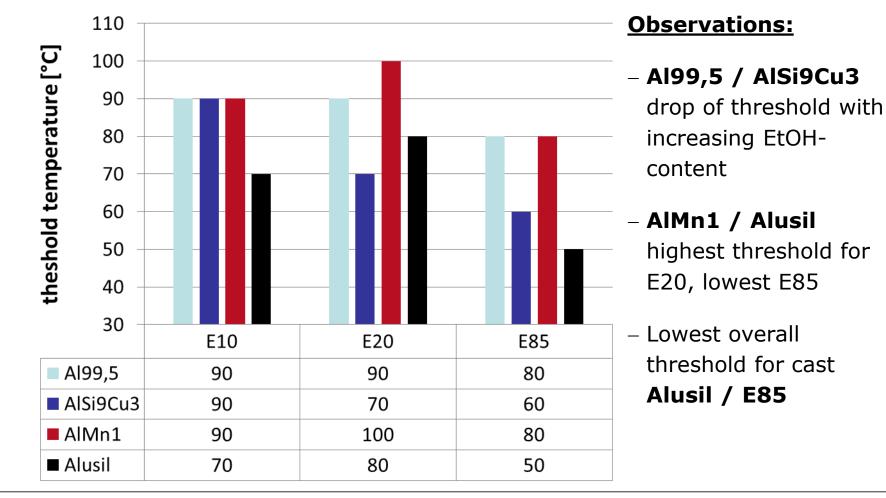






Influence of EtOH-content

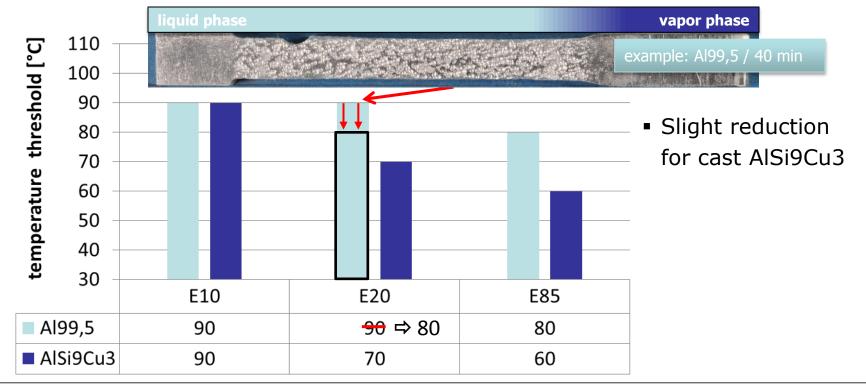






Influence of increased testing time, surface/volume ratio

- Increased surface area: 1 → 4 specimens in one test vessel
- Increased testing time: 168 → 336 h (2 weeks)
- ⇒ Slight reduction of temperature threshold





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Influence of water-content



Starting temperature:

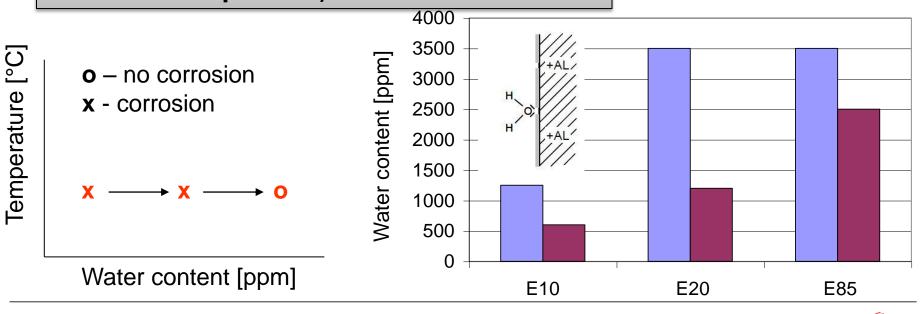
10 °C above the highest temperature level with **assured** alcoholate corrosion → water addition, until no alcoholate corrosion occurs at this temperature level.

⇒ Inhibition of alcoholate corrosion by water

additions is possible, but....

Starting temperature:

Al99,5	AlSi9Cu3
E10: 130°C	E10: 120°C
E20: 130°C	E20: 110°C
E85: 120°C	E85: 100°C



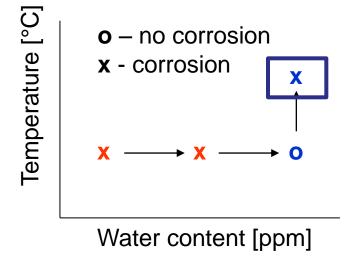


Influence of water-content



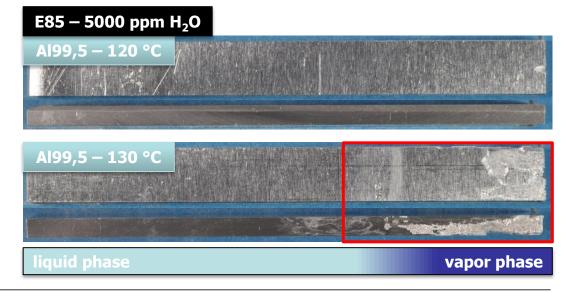
Increase of temperature for the water-inhibited fuel

⇒ Alcoholate corrosion!



Observations:

- At very high temperatures alcoholate corrosion is shifted towards the vapour phase!
- ⇒ At lower temperatures alcoholate corrosion occurs in the liquid phase due to the lowest boiling point of water-rich EtOH-azeotrope



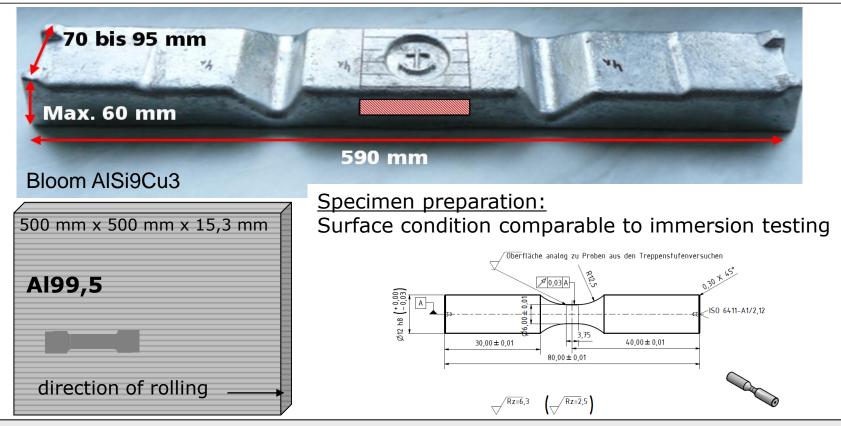


Raw material for corrosion fatigue tests (Stress ratio R = 0, Fuel: E85)



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⇒ Test temperature: 10 °C below static (immersion) temperature threshold
⇒ Load level (R = 0): Stress amplitude lowered by 5 % in relation to the run out level at 1E7 cycles in air

Corrosion fatigue tests

Observations

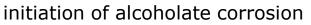


Al99,5 in E85 @ 70 °C:

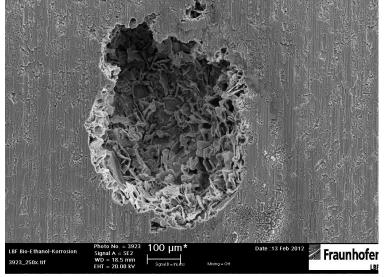
- Fatigue life in E85/70°C reduced by a factor of 10 compared to air
- Signs of initiation of alcoholate corrosion on the fracture surface <u>after</u> rupture

AlSi9Cu3 in E85 @ 50 °C:

 No significant reduction of fatigue life (due to high porosity of the cast material crack initiation still occurs at intrinsic defects), but...



AlSi9Cu3 in E85 @ 50°C - 47,5MPa



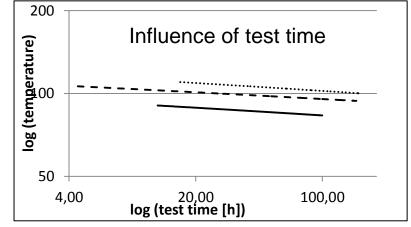


Conclusion

Technological temperature threshold as indicator of susceptibility to alcoholate corrosion



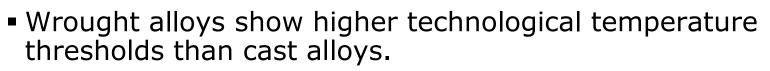
- For a given material an fuel, the probability of alcoholate corrosion is strongly influenced by
 - Temperature (T), water content (W), surface area/volume ratio (A), time (t)
 - Order of impact: T >> W > A > t
- The deduction of a technological temperature threshold as an indicator of susceptibility to alcoholate corrosion is suitable to describe further influences like
 - Manufacturing, surface condition, ...
 - Mechanical loads
 - Size of sample surface
- Threshold can be used to rank technological influences, but not as a design criterion!



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Conclusion

Alcoholate corrosion of aluminium in ethanol blended fuels



- Inhibiting water content is not only dependent on the ethanol content
- The technological temperature threshold is lowered by superimposed cyclic mechanical loading for the cast alloy AlSi9Cu3 (50 °C fuel temperature)
- Corrosive attacks were observed for Al99,5 at 70 °C but could not clearly be assigned to alcoholate corrosion
- Fatigue life of Al99,5 was considerably decreased by the influence of corrosion
- Fatigue loaded components exposed to ethanol blended fuels should be qualified thoroughly by additional corrosion fatigue tests



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