Development and Validation of an Ultra Performance Liquid Chromatography Method for Venlafaxine Hydrochloride in Bulk and Capsule Dosage Form

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Chhalotiya, et al.: UPLC Estimation of Venlafaxine Hydrochloride

A simple, specific, accurate, and precise ultra performance liquid chromatographic method was developed and validated for the estimation of venlafaxine hydrochloride in tablet dosage forms. A acuity TM BEH column having C18, 100×2.1 mm i.d. in isocratic mode, with mobile phase containing dipotassium hydrogen phosphate: Acetonitrile (30:70 v/v; pH 7.00 with dilute o-phosphoric acid) was used. The flow rate was 0.75 ml/min and effluents were monitored at 227 nm. The retention time of venlafaxine hydrochloride was 0.548 min. The method was validated for specificity, linearity, accuracy, precision, limit of quantification, limit of detection, robustness and solution stability. Limit of detection and limit of quantification for estimation of venlafaxine hydrochloride were found to be 6.11 µg/ml and 20.33 µg/ml, respectively. Recoveries of venlafaxine hydrochloride in tablet formulations were found to be in the range of 99.3-99.5%. Proposed method was successfully applied for the quantitative determination of venlafaxine hydrochloride in tablet dosage forms.

Key words: UPLC, validation, venlafaxine hydrochloride

Venlafaxine is a bicyclic antidepressant, and is usually categorized as a serotonin-norepinephrine reuptake inhibitor (SNRI), but it has also been referred to as a serotonin-norepinephrine-dopamine reuptake inhibitor. Venlafaxine hydrochloride chemically is (R/S)-1-[2-(dimethylamino)-1-(4-methoxyphenyl) ethyl] cyclohexanol hydrochloride or (±)-1-[α-(dimethylamino)methyl] p-methoxybenzyl] cyclohexanol hydrochloride salt and has the empirical formula of C_{17}H_{27}NO_{2}.HCL.

Various methods have been reported for estimation of venlafaxine hydrochloride in biological matrices such as plasma, which includes the use of LC with UV detection[1], LC with electrospray ionization mass spectrometry[2], LC with coulometric detection[3], LC with fluorimetric detection[4,5], LC with diode array detection[6,7], GC-MS[8], LC-MS-MS[9] and for estimation of it in serum by use of LC[10]. Stability indicating methods have also been reported for its in vitro determination in gastric and intestinal fluids[11] and pharmaceutical formulations[12]. RP-HPLC method is also reported for estimation of it in tablet dosage form[13].

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The reported stability indicating methods and RP-HPLC uses acetonitrile and buffer in various proportions for quantification of venlafaxine hydrochloride. Present study involves development of UPLC method using simple mobile phase containing acetonitrile and buffer for quantitative estimation of venlafaxine hydrochloride in capsule dosage forms which is sensitive and requires shorter analysis time. The developed method was validated as per ICH guidelines[14,15].

An Acquity TM UPLC Instrument Control Option Pack Version 1.23b system (waters, equipped with a diode array detector, isocratic pump, column oven) rhenodyne valve with 5 µl fixed loop, and acquity TM BEH column having C18 (100×2.1 mm id, 0.2 µm particle size) was used.

Analytically pure powder venlafaxine hydrochloride was procured as gratis samples from Cadila Healthcare Limited, Moraiya, India. HPLC grade acetonitrile and orthophosphoric acid were purchased from E. Merck (Mumbai, India) and were of analytical grade. HPLC grade water was purchased from Milli Q. The marketed formulation containing 150 mg of venlaflaxine (Efflexor-XR, Zydus-Cadila Healthcare Ltd., Ahmedabad, India) were purchased form pharmacy.

The Acquity TM BEH column C18 was used at ambient temperature. The mobile phase consisted of dipotassium hydrogen phosphate: Acetonitrile (30:70 v/v; pH 7.00 with dilute o-phosphoric acid) was used. The flow rate was 0.75 ml/min and effluents were monitored at 227 nm. The mobile phase was passed through HVLP 0.22 µm membrane filter and degassed before use. The elution was monitored at 227 nm, and the injection volume was 5 µl.

UPLC method depends upon the nature of the sample (ionic or ionizable or neutral molecule), its molecular weight and solubility. UPLC was selected for the initial separation because of its simplicity and suitability. To optimize the chromatographic conditions the effect of chromatographic variables such as mobile phase, pH, flow rate and solvent ratio were studied. The resulting chromatograms were recorded and the chromatographic parameters such as capacity factor, asymmetric factor, and resolution and column efficiency were calculated. The condition that gave the best resolution, symmetry and capacity factor was selected for estimation.

Accurately weighed 56 mg of venlafaxine HCl was transferred to a 50 ml volumetric flask and dissolved in 20 ml of the solvent. The flask was sonicated for 15 min. and volume was made up to the mark with the solvent. From this stock solution working standard solution was prepared by transferring further 5.0 ml to 50 ml volumetric flask and the solvent was added up to the mark to give a solution containing 112 µg/ml venlafaxine HCl.

Average weight of the capsule pallets was computed by accurately weighing 20 capsule pallets. Powder equivalent to 100 mg of venlafaxine HCl was weighted accurately and transferred to a 200 ml volumetric flask. After addition of 100 ml of the solvent flask was sonicated for 30 min. Solution was filtered through 0.22 µm HVLP filter 5.0 ml of filtrate was pipette out to 25 ml volumetric flask and solvent was added up to mark to get final concentration of 100 µg/ml of venlafaxine HCl. The prepared sample solutions were chromatographed for 5 min using mobile phase at a flow rate of 0.75 ml/min.

Appropriate volumes of aliquots from standard venlafaxine HCl stock solution were transferred to different volumetric flasks. The volume was adjusted to the mark with the solvent to obtain the concentration of 28, 56, 90, 112, 135, 168 and 196 µg/ml. Area of each solution were measured at 227 nm and the plot of absorbance v/s concentration was plotted. The straight-line equation was determined from calibration curve.

Accuracy is the closeness of the test results obtained by the method to the true value. Accuracy is performed at three levels 50, 100 and 150%. Percentage recovery and low relative standard deviation value shows accuracy of the UPLC method. The precision of analytical method is the degree of agreement among individual test results when the method is applied repeatedly to multiple samplings of homogeneous samples. The relative standard deviation for six replicates of sample solution was less than 2.0%, which met the acceptance criteria established for UPLC method. Ruggedness test was determined between two different days, analysts and instruments. The value of RSD found to be below 2.0% showed ruggedness of developed UPLC method. The linearity of analytical method is its ability to elicit test results that are directly proportional to the concentration of analyte in sample within the given range.
of the analytical method is the interval between the upper and lower levels of analyte that have been demonstrated to be determined within a suitable level of accuracy and linearity using the method as written.

Specificity is a procedure to detect quantitatively the analyte in the presence of components that may be expected to be present in the sample matrix. Selectivity is a procedure to detect qualitatively the analyte in presence of components that may be expected to be present in the sample matrix. Commonly used excipients in capsule preparation were spiked in a preweighed quantity of drug and then absorbance was measured and calculations done to determine the quantity of the drug. The method was found to be robust, as small but deliberate changes in the method parameters have no detrimental effect on the method performance as shown in. Low value of relative standard deviation was indicating that the method was robust. Standard and sample solution stability was evaluated at room temperature for 48 h. The relative standard deviation was found below 2.0%. It showed that both standard and sample solution were stable up to 48 h at room temperature.

Several mobile phases were tried to resolve venlafaxine HCl but the resolution was not satisfactory. So modification was made in the above mobile phase. Finally the system containing acetonitrile: Buffer (70:30, pH=7.0) was found to be satisfactory and gave well resolved peak for venlafaxine HCl. The retention time for venlafaxine HCl was 0.548 min (fig. 1). For the selection of detection wavelength, the spectrum of 112 ppm venlafaxine HCl revealed that, at 227 nm the drug possess significant absorbance. So considering above fact, 227 nm was selected as a detection wavelength for estimation of venlafaxine HCl using UPLC. For selection of appropriate column and flow rate, zorbex column and various flow rates were tried but the satisfactory resolution was obtained with column-aquity and flow rate 0.75 ml/min. The retention time for venlafaxine HCl was found to be 0.548 min and theoretical plates observed were 5705. From the peak area obtained in the chromatogram, the amount of the drug was calculated as 99.3±0.2 % w/w.

The calibration curve for venlafaxine HCl was prepared by plotting area v/s concentration. The

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<th>TABLE 1: DATA DERIVED FROM ACCURACY OF VENLAFAXINE HYDROCHLORIDE THE PROPOSED UPLC METHODS</th>
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<td>Levels</td>
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<th>TABLE 2: SUMMARY OF VALIDATION PARAMETERS FOR VENLAFAXINE HYDROCHLORIDE THE PROPOSED UPLC METHODS</th>
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<td>Paramete</td>
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<td>Correlation co-efficient (r)</td>
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<td>Residual std. Regression (σ)</td>
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<td>Slope of Regression (S)</td>
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<td>LOD (µg/ml)</td>
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<td>Repeatability, (% Assay, n = 6)</td>
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aLOD= Limit of detection, bLOQ= Limit of quantitation
Linearity plot of venlafaxine HCl is found linear with the linear equation $y = 1.324e^{0.044}x + 1.24e^{0.004}$ and correlation coefficient 0.9993 for venlafaxine HCl. This indicates that the method is linear in the specified range for the analysis of venlafaxine HCl in dosage form. The LOD for venlafaxine HCl was found to be 6.11 μg/ml. The method was found to be accurate with % recovery 99.3 – 99.5% and was found within the acceptance criteria (i.e. 98.0 to 102.0%) with acceptable %RSD of not more than 2% at each level (Table 1).

Precision was calculated as repeatability and intraday and interday variation for venlafaxine HCl. The method was found to be precise with CV=0.6 for intraday (n=3) and CV =0.7 for interday (n=3) for venlafaxine HCl. The low CV (i.e. NMT 2%) has observed for the six assay results (Table 2) hence it concluded that the method is precise for the analysis of venlafaxine HCl in their dosage form.

Ruggenedness test was determined between two different days, analysts and instruments. The value of RSD was to be found 0.9 which is within acceptance criteria of below 2.0% showed ruggedness of developed UPLC method. There is no interference of mobile phase, solvent and placebo with the analyte peak and also the peak purity of analyte peak is greater than 990 which indicate that the method is specific for the analysis of venlafaxine HCl in their dosage form.

The method was found to be robust, as small but deliberate changes in the method parameters have no detrimental effect on the method performance as shown in Table 3. The low value of relative standard deviation was indicating that the method was robust. Standard and sample solution stability was evaluated at room temperature for 48 h. The relative standard deviation was found below 2.0%. It showed that both standard and sample solution were stable up to 48 h at room temperature.

Proposed study describes new UPLC method using simple mobile phase for the estimation of venlafaxine hydrochloride in capsule formulations. The method was validated and found to be simple, sensitive, accurate and precise. Percentage of recovery shows that the method is free from interference of the excipients used in the formulation. Therefore the proposed method can be used for routine analysis for estimation of venlafaxine hydrochloride in its capsule formulations.

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In Vitro Antibacterial and Antifungal Properties of Aqueous and Non-Aqueous Frond Extracts of Psilotum nudum, Nephrolepis biserrata and Nephrolepis cordifolia

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Rani, et al.: Antimicrobials from Pteridophytes

Plants are an important source of neutraceuticals that have proved to be effective against important microbial infections of humans. Lower plants are gaining importance in this regard. The present study is aimed at investigating the antimicrobial properties of three selected ferns, Psilotum nudum, Nephrolepis biserrata and Nephrolepis cordifolia.

The aerial parts of the selected ferns, P. nudum, N. biserrata and N. cordifolia, were fractionated in different solvents. These fractions were concentrated to obtain a powder and were tested against nine bacterial and three fungal strains according to disc diffusion method. The water and ethanol fractions were active against most of the tested bacterial and fungal strains, some of these were more effective than the controls tested. Present study suggests that the pteridophytes, P. nudum, N. biserrata and N. cordifolia could be good source of antimicrobials. These natural compounds might be more effective as the microbes may have lesser chance of developing resistant mutants.

Key words: Antibacterial, antifungal, dermatophytes, disc diffusion assay, Nephrolepis biserrata, Nephrolepis cordifolia, Psilotum nudum, pathogenic bacteria

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Use of antibiotics is often associated with adverse effects such as hypersensitivity, depletion of beneficial gut and mucosal microorganisms, immune-suppression and allergic reactions in the user. Besides, the indiscriminate use of antimicrobial drugs has created immense clinical problems in the treatment of infectious diseases, as the microorganisms develop resistance to often used antibiotics. This necessitates development of alternative antimicrobial drugs, for the treatment of infectious diseases, from the medicinal herbs which are a rich source of novel antibacterial

lichens, and actinomycetes, which are available in the environment. These natural products have a number of advantages compared to synthetic antibiotics, such as fewer side effects, lower toxicity, a lower cost, and the potential for the development of new classes of drugs.

The present study was undertaken to investigate the potential of pteridophytes as a source of natural antimicrobials.


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