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Durable Wooden Water Pipes.

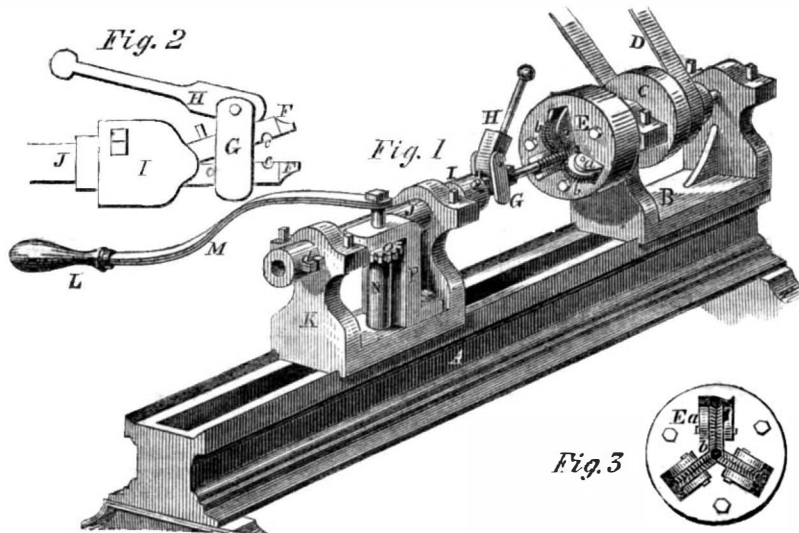
Some wooden pipes laid down for conducting water at Springfield, Mass., by Charles Stearns, Esq., appear to demonstrate the fact that they are more durable in certain situations than pipes made of lead. This plan is to lay them at such a depth as to prevent atmospheric action upon them. In sandy or porous earth, he lays them six feet deep; in compact soil four feet deep, and in peaty or swampy soil three feet deep. In one place heavy lead pipe was laid through a wet meadow, and it required repairs in four years, and had to be lifted in ten. It was replaced by wooden pipes which have now been twenty years in use, and are in good condition yet. The aqueduct pipes which supply Springfield with water have been in use fourteen years, and are still in good order. They are bored logs, the opening being seven inches in diameter, and charred on the inside surfaces by forcing flame through them. The charring of the surfaces of wooden pipes or boards has a wonderful effect in preserving them from decomposition.

It is undoubtedly true that timber sunk deep beneath the surface of the earth, and kept from contact with the air, endures for centuries. We have seen an oak log taken from the bed of a river, in which place it must have remained for hundreds of years, owing to the depth of sand which covered it, and yet it was as fresh as when first submerged. Cedar logs taken from the Jersey swamps, in which they have reposed for a thousand years, are found to be fresh and strong. Wooden pipes are cheaper than those of metal, and are preferable if they can be rendered as durable.

Ginning and Spinning Cotton.

In a letter received from G. S. Yerger, Esq., of Jackson, Mississippi, he informs us that he has put up on his plantation one of Henry's machines for spinning the cotton as it comes from the gin, and that it has been eminently successful. It has now been in operation for two months, and spins No. 5 and No. 10 yarns equal to any in the United States. This combination of the spinning frame with the cotton gin on plantations, whereby the cotton is made into yarn in the ginhouse, saves the expense of packing the cotton, and in the state of yarn transports it to market in a more compact form. If this method of operating cotton is found to be profitable on plantations, we shall soon see it extended so as to embrace the weaving of coarse cotton goods also for the market. This is simply a question of economy with planters. It has appeared to us that it would not be economical unless upon large plantations, where the machinery could be kept running the year round, because we know that machinery in factories deteriorates nearly as rapidly when standing idle as when running.

EVART'S SCREW CUTTER.



The utility of the screw is never to be too highly estimated, as it enters into the construction of nearly everything we use, and is a most essential aid in the manufacture of all machinery. For wood-work they are almost indispensable, and from the vast number which are used, any process or invention which can lessen the cost of their production must in the end, prove a public boon. Such an invention is the subject of our engraving, Fig. 1 being a perspective view of the machine.

A is the bed, exactly the same as a common lathe, on this slides a head, B, capable of being secured in any desired position by a clamp and screw underneath. In this head is a belt pulley, C, receiving motion from a belt, D, and on the mandrel, the cutter, E, is screwed in the position usually occupied by a chuck. This cutter is a circular block of metal, having three grooves in it, placed radially from its center, and it can be adapted to any hand or power lathe. In the grooves slide bearings, a, that can be brought nearer together or further apart by screws, to accommodate any (within certain limits) sized bar. These bearings support cutter wheels, b, of cast steel, which can freely rotate on their axes, and they have angular grooves cut on their peripheries, so that when the three meet together they exactly form one thread of a screw.

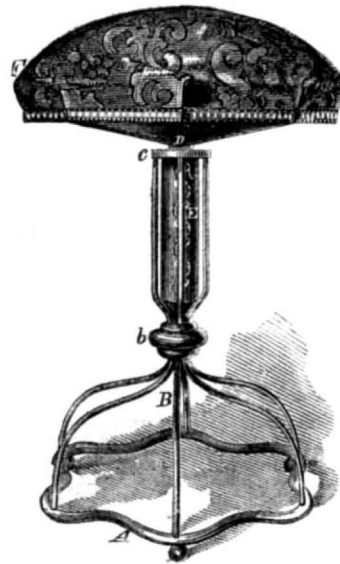
The rotation of these cutters enables them to last much longer than the ordinary ones, and as they are easily replaced, here lies the economy, namely, in the substitution of rotary cutters for the common dies, which are soon injured by the hard scale on the rod, and have to be frequently replaced; and moreover, when the screw is cut it can be immediately withdrawn without stopping the machine, thus effecting a great saving of time.

The screw is placed between a pair of jaws, F, seen enlarged in Fig. 2, and held tight, with its head in the groove, c, by the link and clamp, G H. These jaws are secured in a piece, I, attached to a sliding bar, J, that can move freely back and forth in the head, K. The screw blank being fastened in the jaws, it is fed between the cutters, which are rotating, by the operator pulling the handle, L. This handle is at the end of a lever, M, attached to a small upright shaft, N, that carries a cog wheel, O, and supported in a suitable bearing, P, on the head, K. The cog wheel, O, gears into teeth, d, on the sliding bar, J, and so when L is pulled, J is

moved in its bearings, and the screw fed to be cut. When it is cut, by operating L in a reverse direction, the screw can be withdrawn, and another blank placed in the jaws. Fig. 3 is a front view of the cutter, E.

This valuable machine is the invention of James M. Evarts, of Westville, Conn. Any further information can be obtained from him, or M. Merriman, Jr., & Son, of the same place. The rights for all the States except Connecticut are for sale. It was patented June 16, 1857. Mr. Evarts had this operating machine on exhibition at the Crystal Palace, from which our engraving was taken. The machine was, of course, destroyed by the late fire.

Leach's Music Stool.



The wooden stools usually accompanying a piano are, as every one knows, expensive, and far from being a firm and secure seat, therefore to provide one which is at once light, elegant, cheap, and strong, this inventor—Edwin Leach, of Norwich, Conn.—has produced the subject of our engraving. An elegantly shaped iron rim, A, stands on three knobs or castors, and from it rise six or more light wrought iron rods, B, which are bound together by a hub, b, and from it they again slightly expand, and then pass quite straight to the piece, c, in which they are firmly secured. This piece, c, is a nut in which the screw, D, that supports the seat, F, works, so that by turning the seat round it can be raised or lowered to suit the comfort of the person

who is about to sit down. The screw, D, when between the rods, B, is covered with an ornamented case, E, so that the appearance of the stool is always graceful, tasty and convenient. They can be made much cheaper than wooden ones, and are in every respect preferable, as being capable of enduring infinitely more wear, and present an elegant piece of furniture as an accompaniment to a musical instrument.

It was patented August 24, 1858, and the inventor will be happy to furnish any desired information upon being addressed as above.

Glycerine.

A correspondent asks us if there is any cheap process of obtaining glycerine, asserting it would reduce the price of soap, by turning the soapmakers' waste to some account. There is such a process in extensive operation at Price's candle works in London, England, and 't is so simple and cheap that any soapmaker may put into practice. A continuous current of steam of 600° Fah. is led into a distillatory arrangement containing neutral glycerine fat, and in due time produces the decomposition of the latter into fatty acids and oxyd of glycerine, which distil over in combination with their constitutional water. The glycerine, from its greater density, forms the lower stratum of the distillate, and therefore may be easily separated from the supernatant fatty acids. In this state it is very dilute, and must be concentrated by evaporation until it reaches a specific gravity of 1.240 at 60° Fah., when it is ready for the market.

Shaving Soap Powder.

Most of the soaps in use for shaving may with justice be found fault with. They either do not lather freely, or else they excite an unpleasant sensation, arising from an excess of caustic alkali used in their manufacture. The alkali acts upon the skin as well as upon the beard; and to obviate these inconveniences, or at least to mitigate them, the following process has been invented:—Take about a quarter of a pound of the finest white Windsor soap, cut it into pieces the size of a walnut, place them in a dry and warm situation for several days, until perfectly hard. Now grate the soap up to powder with a nutmeg grater. Place the soap powder in a shaving dish, and pour over it just as much alcohol as will cover it; next day it will be fit for use. Thus prepared, the soap has lost all action hurtful to the skin, and has acquired a remarkable mildness and unctuousness. Instead of plain spirits of wine, any perfumed spirit, such as Hungary water, adds the charm of fragrance. Brown soap does not answer so well as white, because all the brown soaps are of commoner quality than the white, and are artificially colored with burnt umber, &c., which is not only dirty on the towels, but is used purposely to hide by its color other imperfections in the soap.

SEPTIMUS PIESSE.

African Cotton.

Recent intelligence has been received from Dr. Livingston's expedition up the river Zambesi, in Africa. They are now going up that river slowly, and have discovered a peculiar kind of cotton growing in a deserted garden. Its staple is longer than Angola cotton, and the seed does not adhere to it like that of American short staple. A sample of this cotton has been received in Manchester, England; it is clean, and looks very well, but is not to be compared to even the middling quality of South Carolina cotton.

New Inventions.

Rapid Sawing.

At the Penn Mills, near Clarion, Pa., no less than 21,072 feet of boards was recently turned out in ten hours by a circular saw of 4½ feet in diameter. The power employed is a steam engine of excellent workmanship, made made by N. Myers & Bro., of Clarion. It is of 16-inch bore, 20 inches stroke, and makes 150 revolutions per minute (500 feet velocity of piston). Two boilers, each 24 feet long, are employed to generate steam; they are 42 inches in diameter, have two 16-inch flues each, and the fuel used is sawdust and waste slabs.

Safety Chemical Matches.

A Frenchman, named M. A. Meunons, has secured a patent in England for an improvement in lucifer matches, with a view to obviate the risks of accidental ignition. To attain this end, the matches are first cut by known means from cubes of wood, the cut being stopped at a short distance from the end of each cube, so as to leave the lower extremities adherent. The upper or free extremity of each packet of splints thus formed, being coated with wax or sulphur, is dipped in one of the following preparations:—Chlorate of potash, two parts; pulverized charcoal, one part; umber, one part: or, chlorate of potash, sulphur and umber in equal parts, thoroughly mixed with glue. The opposite extremity or "cut" of each packet is then painted over with amorphous phosphorus blended with size, so that on separating the matches, the phosphorus is only found on the top of each. The matches thus prepared are ignited by breaking off a small piece of the phosphorized end and rubbing it on the opposite extremity covered with the inflammable preparation.

New Boiler Alarm.

This is an ingenious and simple device for the purpose of calling attention by a whistle, when the water falls to so low a level as to endanger the bursting of the boiler by the water falling below the tubes, the introduction of cold water then being very liable to produce an explosion.

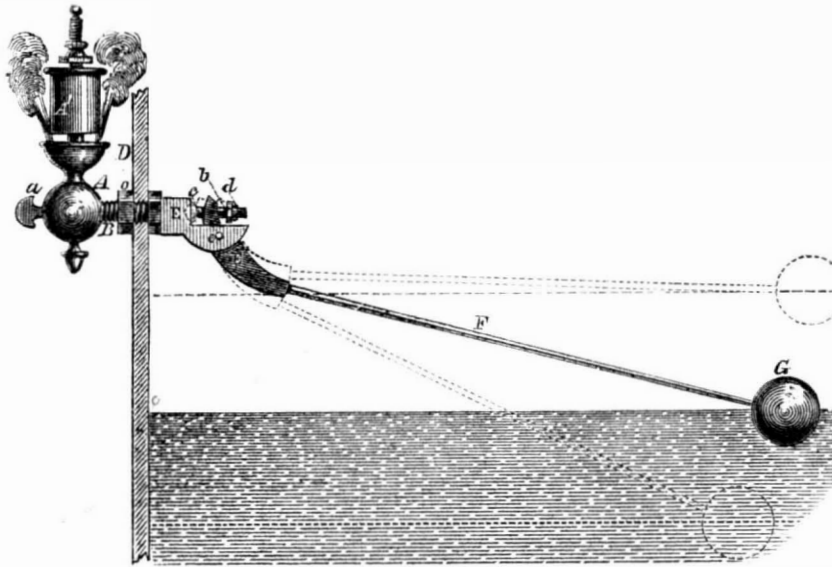
This little apparatus consists of a block of metal, A, to which is attached a whistle, A'. D is the front plate of the boiler, through which the invention is secured by the screw, B, and nut, c. Through the solid part, E, inside the boiler, a passage is bored, corresponding with one in A, and this is flared out to form a valve seat on the inner surface of E, which is closed by a conical valve, c, upon the rod, b, the valve being kept home by the pressure of the steam behind it. The other end of this rod passes through a piece, E', which is hinged by the pivot, e, to E, so that it can turn freely round, and by pressing against the nut, d, on the valve rod, b, bring out the valve, c, from its seat, and allow the steam to pass to the whistle, and call attention to the fact that the water is low. To the lower part of E' a long bar F is attached, having a hollow ball, G, at its end, which floats on the surface of the water.

As the water rises and falls in the boiler within proper and safe limits, the ball rises and falls with the water, moving the piece, E', upon the pivot, e; this piece is so shaped that the moment the water approaches the dangerous point, the weight of G and lever, H, is sufficient to force back E' upon its pivot, e, and pull the valve, c, from its seat, thus allowing the steam a free passage to the whistle, and it will continue so to do until more water is supplied to the boiler, or the defect in the pumps, if there be one, remedied. Much of the water used for boilers contains a great quantity of solid matter, and this is liable to be deposited around, c, and prevent the proper action of the alarm. This has hitherto been a great defect in boiler

alarms, but in this one it is provided against, by the addition of the screw, a, outside the boiler, so that when the stoker suspects that there is any matter collected at c, or at regular intervals, he can, by moving the screw, a,

force c back, and either grind it in its seat, or permit the steam to blow the solid matter away. The ball and lever in the positions indicated by the dotted lines show the range which they have before operating the valve,

MILLER'S BOILER ALARM.

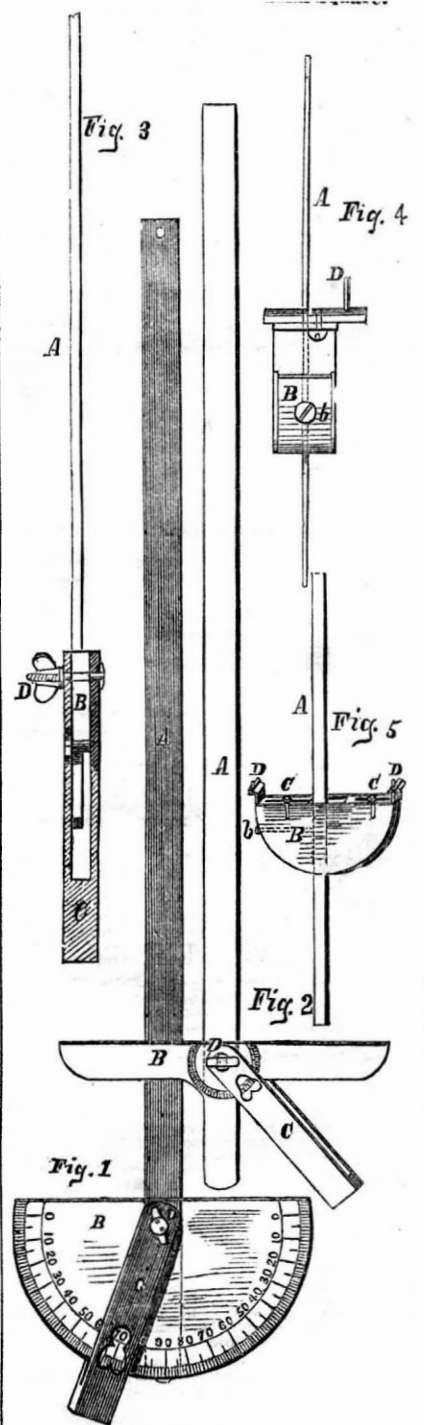


the lowest position being that in which the ball allows steam to escape and produce a whistle.

This most valuable device, which is also an ornament to a boiler, is the invention of Alexander Miller, of Cleveland, Ohio, and was

patented by him June 29, 1858. Any more particulars that may be desired can be obtained by addressing the inventor, care of Dr. Seelye, at the above place. We have seen certificates from some Steamboat Inspectors, who speak highly of the invention.

Bronson's Bevel and Radial Square.



These two inventions are designed, the one to improve and extend the use of the T-square

and make it more completely the *vade mecum* of the draughtsman, and the other to provide a simple instrument by which shafts may be centered with greater facility.

Fig. 1 is a front view of the bevel square. It consists of a straight rule, A, and a semi-circular head, B, the lower part of which projects slightly below A like a common square, so that it can be put against the side of a drawing board to rule horizontal lines along the side of A. A pin and nut, D, secures the piece, C, to the square; this piece has a perforation at its end, and a point which allows it to be placed at any angle, and at which it can be secured by tightening the nut, D. This piece, C, being rested against the side of the drawing board, a line at the same angle can be drawn on the paper, guiding the pencil by A, and by turning the square round to the other side of the board a line can be drawn at the same angle in a reverse position. Fig. 2 shows another square on the same principle without the extended semi-circle, and with the graduations on the point where A and B join. Fig. 3 is a side view of the same.

Fig. 4 is a side view, and Fig. 5 a front view of the radial square, which is as simple as anything for the same purpose could possibly be. It is often necessary to find the center of the end of a round shaft or piece of wood to "center," as it is called, for turning, and this little instrument does it immediately. A block of wood, B, bound with brass for protection, has a piece of metal, A, passing through it, and A can be fixed in any position by a screw, b. Two small projecting pieces, D, are attached, so that they can slide to or from A, and be secured in any position by screws, C.

The method of using is very simple. These pieces, D, act as callipers, and being properly adjusted in relation to A, they are placed on the outside of the shaft with the flat part of A resting against the end of the shaft, a line is then drawn along A, and the whole turned into another position, the pieces, D, still being pressed against the periphery, and A still being kept flat against the end of the shaft, another line is then drawn, and the point of intersection is the center of the shaft. Its principal use, however, is the marking out the teeth of bevel gearing.

These two convenient instruments are the invention of Austin Bronson, No. 102 Elm st., New York, who may be addressed for further information.

Method of Preventing Seasickness.

Of all the ills that human flesh is heir to, there is none so nauseating and thoroughly odious as seasickness, and we have no doubt that all of our readers who have ever been afflicted with it, and again contemplate "going down to the sea in ships," will hail a preventive as a boon more highly prized than a princely diadem. An alleged preventive for seasickness, and illness arising from similar causes, has recently been patented in England by an Italian residing in France, named P. Molinari, which consists in the use of a composition prepared in the following manner:—First, soak in a pint and three-quarters of vinegar for about twelve hours, rue, ½ oz.; turmeric, ½ oz.; green husks of walnuts, ½ oz.; rocon (annatto), ½ oz.; potash, ½ oz., and a poppy head. Then boil the whole for half an hour, and strain through fine linen. Pieces of filtering paper in four or five thicknesses, measuring about ten by seven inches, are to be soaked in the solution, and when dry, sewn around the edges to pieces of some light fabric, wadding being placed between the paper and fabric (or the wadding itself may be soaked in the solution), and tapes attached to the fabric, to attach it to the person. By applying pieces of paper or wadding prepared in the above manner to the pit of the stomach, it is assumed that seasickness, and illness of an analogous character, such as that caused by the shaking of railroad cars and other carriages, will be prevented.

Wearing Apparel.

The London *Medical Times* contains an article on the above subject by Dr. Collier, who has been investigating scientifically the nature of different habiliments as agents for protecting soldiers against high heat. By placing a thin layer of white cotton over a soldier's red woollen cloth coat, exposed to the sun in India, a fall of seven degrees in its temperature soon took place, hence he recommends that the colored clothing of soldiers should be covered with white cotton cloth when they are marching in the hot sunshine. All kinds of clothing he found were capable of absorbing a quantity of moisture from the body. Woolen cloth absorbs the greatest amount, and cotton the least. From this we should conclude that cotton flannel was better than woolen flannel for under garments, an opinion quite contrary to the one generally entertained. The color of clothing has very little sensible influence in reference to the heat of the body, leaving solar heat out of the question. Black, white, red, blue and brown clothes are equally warm, their composition and texture being equal in all other respects.

Cost of Electric Light.

M. Edmond Becquerel, a French *savant*, has been recently engaged in some experiments with a view to determine the comparative cost of electricity as an illuminating agent. He used a battery of zinc and platinum, made with strict attention to economy, and the results were as follows:—

The standard is the light of 350 candles of the best quality, and the cost of
 Coal gas, at \$1 60 per 1000 c. feet, was \$0 35
 Oil (Rape seed), at 17 cents per lb..... 0 60
 Stearine candles, at 32 cents per lb..... 2 52
 Wax candles, at 52 cents per lb..... 3 12
 Electric light..... 0 58

Thus showing that although the electric light is cheaper than candles, it will not at present compete with coal gas, at least until some cheaper battery power be found.

LAC VARNISH.—This is made by dissolving gum lac in alcohol, or in a solution of the carbonate of soda or potash. It is easily made, and is used for many purposes. Oil varnishes are the best for coating iron utensils exposed to the weather, especially the sulphur varnish, which is exceedingly useful for farmers. These varnishes may be made of different colors by adding pigments. Verditer blue is mixed with copal varnish for polished iron work that is to be exposed to the atmosphere.

Scientific American.

NEW YORK, OCTOBER 16, 1858.

The Crystal Palace in Ruins—Fair of the American Institute Closed.

The Crystal Palace, which has been so much admired for its architectural beauty and a source of so much litigation for the past five years, is now a mass of ruins! It was totally destroyed by fire—lighted probably by the torch of an incendiary—on Tuesday, the 5th instant, between the hours of 5 and 6 o'clock in the afternoon.

This structure was erected in 1853, for the purpose of an exhibition of the industry of all nations. From the fact of so many of our citizens having visited London during the World's Fair in 1851, and witnessing with some regret the limited display of American art on that occasion, they were led to take steps for an exhibition in this city, where such a display could be made as would do justice to American skill and genius. In March, 1852, the Legislature of New York granted a charter, and the city of New York granted a lease of Reservoir-square to an association for carrying out these objects. Theodore Sedgwick, Esq., was its first President, and it was incorporated with a capital of \$300,000. Measures were early taken for the erection of a proper building, and among the several designs sent in for this purpose, that of Messrs. Carstensen & Gildemeister was selected and carried out into execution, C. E. Detmold being superintendent and engineer. We published illustrations and full descriptions of the building in Volume VIII. of the SCIENTIFIC AMERICAN, but its cost far exceeded all early expectations, amounting to more than \$700,000. It was to be opened in May, 1853, but it was not ready in time, and the inauguration was postponed till the month of July following. On that occasion President Pierce was present and graced the assembly. National Commissioners from all the European governments took part in the ceremonies, and the display was grand in every respect. The exhibition which followed was a good one—the best ever held in our country—but it turned out an unprofitable enterprise for the stockholders, a dividend never having been declared from the time it first went into operation until the building fell into the possession of the city authorities.

In about nine months after the Fair opened, the finances became very embarrassed, and Mr. Sedgwick resigned. In 1854, P. T. Barnum was elected to fill his place, under the idea that he could render it a more successful enterprise, but it also failed under his management. The building and all its contents afterwards passed into the hands of a Receiver, and was occasionally let out for concerts and balls, and to the American Institute for their annual exhibition. The five years' lease of the property upon which the Palace was built having expired, the city authorities, without unnecessary circumlocution, ousted the Receiver and entered possession. Its destruction by fire consummates a series of troubles and vicissitudes of no ordinary character.

The Fair, at the time of the fire, had attained to completeness in its arrangements in some respects superior to all its predecessors. Twenty-five per cent more articles were entered than during the Fair of 1857, and a large number of these were meritorious improvements.

We were engaged in the building all the afternoon of the day on which the fire took place, in examining into the novelty and merits of various machines and articles on exhibition. At five o'clock the face of all things bore the appearance of confident safety; in fifteen minutes afterwards the flames had ascended from basement to roof, and were whirling round in terrible eddies inside of the dome. In twenty minutes after

the fire was first observed, the lofty dome surged and fell with a crash like thunder, and the entire roof soon came down, together with nearly the whole of the outer walls. Never was destruction more complete. The whole space on which the Palace once stood is now but a heap of confused pieces of broken iron, embracing all kinds of machines, from the ponderous steam engine to the watch-maker's lathe.

The heat caused by this fire was very intense, and a whirlwind was thus generated which carried up some of the roofing plates, and bore them to a distance of three miles on Long Island. The glass of the roof was melted like wax, and among the ruins was found a case of artificial teeth, perfectly sound, but enclosed in a ball of the vitrified roofing. There were over three thousand contributions, upon exhibition, few of which were saved.

The fire was so unexpected and sudden that nearly all present were either stupefied or appalled. One act of heroism performed by Sarah Stevens, a factory girl from Manchester, N. H., deserves to be especially mentioned. She was attending the knitting machines noticed by us last week, and succeeded in saving three of these machines by three separate runs into the burning structure, and she carried off the last one when the case on which it stood was in flames. The two brothers Aiken saved other four of these machines, thus making seven in all—a greater number proportionably than was saved by all the other exhibitors.

We cannot find language adequate to express our sympathy for them in the sad affair. Suffice it to say that hundreds of honest and industrious mechanics, manufacturers and inventors have lost valuable property, and had their cherished plans and hopes frustrated. The average loss to exhibitors is estimated to amount to about \$300,000, and if sympathy has any value in such a case, it is extended by the community generally. There is one consideration in connection with this calamity which affords cause for gratitude: no life has been lost. There were more than a thousand persons in the building when the fire broke out, and quite a number of women and children among them; but all got out in safety, although the conflagration was terribly rapid in its results. Had the accident occurred two hours later in the evening, when several thousands would, in all probability, have been thronging the building, hundreds of lives might have been sacrificed.

Fires on Land and Sea—What Can be Done to Suppress them?

We have on so many occasions directed attention to the best means of preventing and extinguishing fires in ships and buildings, that it almost seems like repeating an old story to recur to the subject again. But while fires of a most disastrous character are still frequently taking place, we feel called upon to allude to them; and will endeavor to present such information and considerations as are applicable in such circumstances. It is a self-evident fact to most minds that carelessness and the absence of provident forethought are the causes of a majority of the accidents which occur; but these considerations can afford no reason why we should not urge the most efficient preparations for such contingencies. On board of the ill-fated steamer *Austria*, neither forethought for preventing fire was displayed, nor were there any sensible means prepared to meet such an emergency. There was one fatal error connected with the management of that vessel, which seems to have been overlooked by most persons, namely, the unsafe position of the powder magazine. At a very early stage of the fire it exploded, and this, we believe, caused the magnitude of the disaster, by creating universal consternation on board, and suffocating the engineers by "after-damp": hence the subsequent mismanagement and all the well-known dreadful circumstances which ensued. On the recent trip of the steamer *Hammonia*—the *Austria's*

consort—her powder magazine exploded, injuring several persons, and proving uncontestedly, that the magazines of these vessels were located in dangerous positions. Commanders of all vessels are bound to consult safety rather than convenience, in regard to the position of their magazines and every other explosive and combustible material on board.

It appears to be a plausible conclusion, that by having ships and buildings constructed of a fire-proof material, such as iron, it would be a means of preventing fire. But the *Austria* was an iron ship, and the Crystal Palace an iron building; yet neither the hull of the one nor the iron frame of the other prevented their destruction by fire. It is, however, folly to expect that the incombustible materials of which any such structure is composed can prevent combustible materials within from being consumed; but in all such cases the proper course to pursue is to provide all possible means of extinguishing fire.

We are aware that it is no new doctrine that steam can be usefully employed as a fire-extinguisher—not only in steamers, but also in factories and all buildings where boilers are used; yet, such is the almost criminal neglect in not providing to take advantage of it, that we feel called upon to sound the alarm once more.

The Engineer at the Crystal Palace, taking advantage of this agent, and to save the probable explosion of the boilers, promptly threw open their valves, and thus allowed the steam to escape into the burning building. Now, on board of a steamship, this decision of character in like circumstances might save it from destruction; but in a building covering such an extensive area as the Crystal Palace, it produced no useful result in suppressing the conflagration—the quantity of steam being insufficient to fill the exposed parts of the building.

Steam acts as a fire-extinguisher, principally by its greater expansiveness than air, whereby it penetrates through all the space adjacent to the burning materials, and thus excludes the oxygen from supporting combustion. In the rooms of factories, and in the compartments of steamers, sufficient steam may be supplied from the boilers to dispel the air and arrest combustion, in cases of fire; and we are greatly surprised that this means is not provided, as its efficiency is of no dubious import.

On page 52, Vol. VIII., SCIENTIFIC AMERICAN, an account is given of a cotton factory, in Douay, France, which was saved from destruction by the prompt application of an axe to a steam-pipe running through the room in which the fire originated. We have also a letter now before us from Mr. T. Sault, of Seymour, Conn., who, while engineer of a steamer, a few years since, saved it from being consumed, by the same agent. A tin can containing turpentine had been accidentally overturned in the boiler room and having taken fire, the flames spread with fearful rapidity while no water could be brought to bear upon them. There was a small pipe leading from the boiler to the smoke-stack: this the engineer promptly wrenched off, and the steam escaped in volumes, quickly filling up the entire space, and extinguished the flames, as if by magic. As the vessel was built of wood, and the place where the fire took place contained a great amount of combustible materials, its entire destruction would probably have resulted, but for the timely application of the steam jet. On page 368, Vol. IX., SCIENTIFIC AMERICAN, Mr. A. Walker, superintending engineer of the ninth district, recommended that all steamboats should be fitted with iron pipes leading from the boilers to the several apartments, for quenching fires by steam; and although his suggestions were excellent, we are not aware that they have been applied to a single steamboat in our whole country. In 1852, a large British steamship, called the *Amazon*, was consumed by fire on her first voyage. Previous to going to sea, it was proposed to arrange steam-pipes through all her rooms, for extinguishing fire; but this

suggestion was not carried out, although the cost would only have amounted to about \$1000.

In making arrangements to meet dangers, a penny-wise and a pound-foolish policy seems to govern, and we suppose that, however much may be said by the press, the same indifference will still continue; but this shall not deter us from urging every consideration that shall act as a preventive of such terrible disasters. In all steamships and manufactories where steam can be employed for extinguishing fires, means should be arranged to attain this object. Pipes leading from the boiler into every apartment could be provided, each having a valve or cock, in order to concentrate the steam upon the point of pressing danger, and thus made to expend its action in the most scientific manner. We do not, however, advise the use of such means to the exclusion of water, as a most efficient agent for quenching fire. In all vessels and buildings where this is possible, a stationary pipe leading from an efficient pump, should be in every apartment, even where there is also a steam-pipe, as no useful means for critical cases of dangers should ever be neglected. And although we exposed the incompetency of Phillip's Fire Annihilator for extinguishing most of the fires which occur on land, we always admitted that it was a valuable safeguard when used in confined places, like the holds of ships, for arresting fires in an incipient stage.

Our object is to urge the adoption of all known means and measures of safety for preventing and extinguishing fires. In doing this, we are well aware that without decision of character and presence of mind in cases of danger, the most efficient means provided for safety may be rendered nugatory; but this is no argument against the prudent preparation of all such means, but rather in favor of them, because provision made for safety tends to dispel those fears which are so liable to paralyze effort in cases of great danger. We are confident that, with the exercise of provident forethought and the adoption of well-known efficient measures for preventing and extinguishing fires on land and sea, ninety-nine out of every hundred of such disasters may either be prevented or so mitigated as to be rendered comparatively harmless. This is a subject that concerns every one. We intend, in our next number, to present considerations which we deem of great consequence in the saving of life in cases of fire and shipwreck at sea.

English Patents.

It is not an uncommon occurrence that useful American inventions are secured by patent in England to parties who have no moral right to them. The English law does not protect the inventor against the introducer of an invention, hence inventors in this country should, if they intend to take patents in England, use the utmost diligence in securing their rights. As examples in point we may mention that Robjohn's ingenious inkstand—illustrated on page 160, Vol. 13, SCIENTIFIC AMERICAN—has been recently patented in England by a British subject, and is heralded in the papers there as "ingenious, and well calculated to prove a great acquisition to the library table." Kurth's improvement in umbrellas, patented last April, was patented in England in July, by parties other than the inventor. Inventors who wish to consult with us in reference to taking European patents, are assured that we have the very best facilities for its careful prosecution. It is unsafe to trust an English case with any other than an agent of experience and integrity, and who is familiar with the law and practice of the British Patent Office. Our agents in London have had nearly forty years' experience in this business.

We learn that Mr. R. D. Clark, of Pennsylvania, has been appointed Chief-examiner of Class V (Calorifics), at the Patent Office. Mr. Clark was for a short time Assistant-examiner in another class. The appointment is a good one.

Hardening Iron.

Every improvement in the manufacture of iron, which is to us the "King of Metals," is to be hailed by the productive world as a positive blessing; and however slight those improvements may be, they deserve the attention of the chronicler's pen; how much more so, then, when they are important and practical, as are those we are about to mention.

The first is the invention of a French clergyman—Chas. Pauvert, of Targe, France—and consists in purifying iron by chemical means. He places the iron in the cementing furnace with 33 parts by weight of finely powdered charcoal, 33 parts of highly aluminous clay, 33 parts of carbonate of zinc or wood ashes, 1 of carbonate of soda, and 1 of carbonate of potash. This produces an iron which has all the properties of the best steel, and it will not lose any of its properties by being heated or drawn out. These substances by chemical action, when heated together, present the carbon in the best possible state to combine with the iron. The method of producing cast steel from this is by melting it in a crucible with about 5 to 6 per cent of the following mixtures:—4 parts of dry carbonate of soda, 3 parts of dry carbonate of potash, 3 of wood ashes, 2 of borax, 3 of oxyd of manganese, and from 4 to 7 parts of charcoal, or some highly carbonaceous body. The 4 parts of carbonate of potash may be replaced by 2 parts of caustic potash. This produces a steel of superior quality, and with more certainty than by the old method. M. Pauvert patented his invention in this country March 23, 1858.

The next invention is that of an Englishman—G. J. Fanner, of Birmingham, England—which consists in using ferrocyanide of potassium, hydro-chlorate of ammonia, and nitrate of potash in equal proportions. These are reduced to a fine powder and incorporated, and a bath made of the same substance dissolved in cold water, the prussiate of potash two ounces, the sal-ammoniac four ounces, and the salt-peter two ounces to every gallon of water. Having now the powder and the bath, the article to be hardened is heated in an open fire or furnace, and rolled in the dry powder until the surface is covered with a pellicle of fused powder, and then it is plunged in the bath where it is left until cold, and when perfectly cooled the mass is hardened. Large masses can be thus rendered extremely hard, but it seems to us to be especially applicable to the hardening of tools, journal bearings, and the like. This process was patented in the United States, April 6, 1858.

Last, but not least, comes an American-invention, that of Herace Vaughn, of Providence, R. I., and patented by him March 30, 1858. He employs two pounds of bi-chromate of potash, twelve pounds of chloride of sodium, and four pounds of prussiate of potash; these ingredients are powdered and mixed together, and they are placed in an iron box, where they are covered with powdered charcoal, and heated in a proper furnace. The articles to be hardened are then placed in the mixture, and the whole heated until the mixture is in a state of igneous fusion, when they are removed and dipped into water, oil or certain solutions in the usual manner. The proportions for hardening wrought iron are different, being 25 per cent of prussiate of potash, 65 per cent of chloride of sodium, and 10 per cent of bi-chromate of potash; bone ash or animal charcoal or both are then added, and the whole is reduced to a state of igneous fusion, and the articles to be hardened are then put in.

Nearly all the inventions of late for hardening iron have been the result of chemistry, and we think that the more perfectly the chemical changes which occur in the transmutation of iron into steel are understood, the nearer we shall be to that great desideratum, making steel directly from the ore, which is the end to which all improvements in iron manufacture are tending.

Telegraph Conductors.

[The following is the first of two articles on this deeply interesting subject. The author is a thorough practical electrician, whose name stands very high as an inventor and electric engineer.—Eds.]

Messrs. Editors—I propose to compare the Atlantic Telegraph Cable with some other conductors more familiar to telegraph operators, and to explain some terms used in works on electrical science; also to give some simple formulae and constants to enable others to verify my conclusions.

In order to establish a current of electricity, peculiar substances, called "conductors," in which it can flow, must be employed. Suitable substances are also required to be placed in such relationship to one another that there must be a chemical reaction between them. For example, if a plate of zinc and another of copper united together by a copper wire, be dipped into a vessel containing a mixture of one part of sulphuric acid and twelve of water, the zinc will be dissolved by the action of the acidulated water, while the copper will scarcely be acted upon. The chemical actions going on in the galvanic cell while the two plates are thus united, are as follows:—Water is composed of two gases—oxygen and hydrogen. The oxygen of the water in the battery having a greater affinity for the zinc than for the hydrogen with which it is associated, leaves the latter, and unites with the former, producing the oxyd of zinc. The hydrogen makes its escape by way of the copper plate, on the surface of which it appears in bubbles, that pass off after bursting. The copper plate is, therefore, called an "electrode," or electric way. The chemical action described would very soon cease in a battery by the accumulation of oxyd on the surface of the zinc plate, were it not for the presence of the sulphuric acid, which unites with the oxyd, forming sulphate of protoxyde of zinc, commonly called sulphate of zinc. This oxyd is very soluble in water, therefore it is taken up by the water in the battery as soon as formed, thus leaving the surface of the zinc tolerably clean, for the continued action of the oxygen, until the water is saturated with the sulphate. Hydrogen escapes more freely from a rough than from a smooth surface, and for this reason the copper plates of batteries are all roughened. But what is going on in the connecting wire of the zinc and copper plates while the chemical action described has been going on in the battery? If some short sections of the copper wire which connects the two plates be replaced with thin platina, or iron wire, these latter will become red-hot, thus indicating that electricity is flowing through the wire, and that its free passage is resisted in the iron or platina, and this is attended with the evolution of heat.

Again, if we place the two plates about a foot asunder in a glass vessel containing the acidulated solution described, and if we unite them by a wire running north and south—the copper being at the northern end—the electricity is said to flow from the copper to the zinc plate. If we now suspend a magnetic needle above and near the wire, it will not point, as it usually does, to the north, but will declinate to the west. This declination will be great according to the diameter of the wire, the size of the plates, the nearer they are together, and the proximity of the needle to the wire. The declination of the magnet will also be greater if the wire is made of silver or copper than if of lead, iron, or platina; also if the battery is more intense in action, such as by substituting nitric sulphuric acid in the solution. From this it is inferred that chemical actions have much to do with the production of electricity, and that all bodies offer a certain amount of resistance—some more and some less—to the flow of a current of electricity. It is found to be rigorously true that the quantity of electricity flowing in the conductor of an electric circuit is directly proportional to the intensity of the chemical action, and necessarily proportional

to the total amount of resistance offered to its passage through the conductor. The following equation will, perhaps, convey this idea more clearly:—Let E represent the intensity of the chemical action, R the amount of resistance offered by all the conductors through which the current passes, and Q the amount of electricity passing at any given time—then we have $Q = E \div R$. The quantity of electricity or strength of current flowing in the wire at any given time being equal to the intensity of the chemical action in the battery, divided by the amount of resistance.

The phenomena presented by electricity when at rest belongs to the science of electrostatics; the phenomena of electricity in motion belongs to electro-dynamics. The latter must be chiefly considered in the study of the telegraph. Bodies are usually divided into two classes in connection with the flow of electric currents; these are called conductors and non-conductors. These terms are merely relative, as it is difficult to specify the dividing line between them. The best conductors are silver, copper, and gold. Gutta percha, glass, and india rubber are the best non-conductors. As all bodies offer resistance to the passage of an electric current, that body is the best conductor which offers the least, and *vice versa*. The resistance offered by each body is called its "specific resistance." The resistance of copper being considered unity, that of silver is .67; iron, 5.625; mercury, 50; distilled water, 79,000,000. The resistance to the flow of a current of electricity also depends on the dimensions and form, as well as the nature of a conductor; but of this we shall treat in the succeeding article. * * *

Boston, Mass., October, 1858.

Marking Ink for Lines.

Messrs. Editors—Those who intend to prepare marking ink according to the recipe in No. 49, Vol. XIII, would do well to dissolve the nitrate of silver in rain water instead of liquor of ammonia, or else they might expose themselves or others to great danger.

Whenever a salt of silver comes in contact with ammonia, it is apt to form with it a compound more dangerous than the fulminate of silver, because more liable to detonation by the most trifling circumstances, and more terrible in its effects. The ordinary way to produce fulminate of silver is this:—Oxyd of silver is mixed with common liquor of ammonia, and left to itself for several hours, when a black powder, this identical compound, will be formed. Another way (and the one so closely resembling the recipe for the marking ink that my fears were instantly aroused) is this:—To a solution of nitrate of silver add strong liquor of ammonia. The dangerous compound in question being thus formed and in solution, is precipitated as a black powder by an alkaline solution—soda or potash. I would warn your readers not to meddle with this substance at all, for I know cases in which the death of the experimenter was caused by a drop of water falling on the preparation. The rule for guidance, therefore, in experimenting with silver is, keep the ammonia (and its salts) out of its way.

Savannah, Ga., October, 1858. L. K.

A Novel Steamer.

Some of our cotemporaries announce that Messrs. Winans, of Baltimore, Md., have just constructed a steamer which is intended to surpass in swiftness every craft that skims the seas, because she is designed to plow through the waves, and not take the trouble of rising and falling with them. This vessel is 170 feet long, and tapers to a wedge edge at both stem and stern. It is to be driven with one large center wheel having diagonal paddles, and to have very powerful engines in proportion to the tonnage. Diagonal paddles and center wheels are not new, and unless it be in the model of this steamer, we cannot perceive from the descriptions published concerning her, anything novel, or any point on which to base conclusions for greater speed than is attained by other steamers.

Useful Receipts.

OIL VARNISHES.—No. 1.—Every person should know how to make these preparations for rendering objects waterproof. Linseed oil is the best to use for this purpose; but as it dries with some difficulty, and is liable to become sticky, it requires to be treated in such a manner as to partially oxydize it, after which it dries quickly, and forms a most excellent varnish. Take a gallon of pure linseed oil, and boil it over a gentle fire for about one hour, adding gradually four ounces of sugar of lead, or litharge, or the oxyd of manganese, or the sulphate of zinc—any one of these will answer—but they must be added cautiously, and the oil stirred well while the oxyd is being fed in. The clear of this is the varnish, the sediment should be mixed with paint. Silk or cotton cloth receiving several coats of this varnish becomes completely waterproof.

No. 2.—Take a gallon of pure linseed oil, and add to it two ounces of sulphuric acid; stir well, put it over a gentle fire in a proper vessel, and boil it for one hour. When cool it is fit for use. The sulphuric acid renders the oil quick drying, and removes its tacky character. This is a good recipe for painters, and manufacturers of oilcloth.

No. 3.—Take one gallon of linseed oil, and add to it about one pound of the flowers of sulphur, and boil for one hour. This is, perhaps, the best oil, or any other kind of varnish of waterproof coating, for outside work, such as porous stones or bricks which imbibe moisture. It will also render statues or other works of plaster of Paris impervious to moisture, and enable them to stand exposure to the weather. It is an excellent preservative oil varnish, and one of the most simple to make.

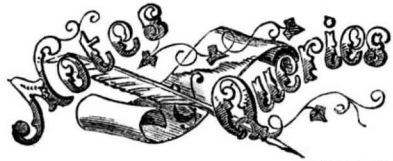
During the time any of these varnishes are boiling, fine shreds of india rubber may be added, and will be dissolved, and render the varnish much thicker, and superior for some purposes. In boiling oil care must be exercised not to allow the fumes to come in contact with flame, or they will take fire. The oil is also liable to fume over when the sugar of lead is added, hence the necessity of stirring well at that particular period. Turpentine renders oil drying, but it also injures, in a great measure, its durable qualities, by imparting to it a partly saponaceous character. The smallest amount possible of turpentine should, therefore, be used in oil varnishes or paints.

COPAL VARNISH.—Take one pound of gum copal, fuse it in an iron vessel over a fire, then add one pint of hot drying oil, like No. 1 varnish, and stir all well until the gum is dissolved among the oil, and the varnish becomes stringy. When cool it is thinned with turpentine. All varnishes improve with age when kept in close vessels. Copal varnish is employed for japanning tinware.

CRYSTAL VARNISH.—This beautiful varnish, which is used for maps, prints, and drawings on paper, is made by dissolving pure Canadian balsam in rectified spirits of turpentine. Equal parts of the balsam and turpentine are mixed together in a bottle or stone-ware vessel, which is set in hot water, and kept in a warm situation for about a week. During this period it should be frequently agitated.

A BLACK JAPAN VARNISH.—Bitumen, 2 ounces; lampblack, 1 ounce; Turkey umber, $\frac{1}{2}$ ounce; acetate of lead, $\frac{1}{2}$ ounce; Venice turpentine, $\frac{1}{2}$ ounce, boiled oil, 12 ounces. Melt the turpentine and oil together, carefully stirring in the rest of the ingredients, previously powdered. Simmer all together for ten minutes.

HAIR LOTION.—Take one pint of alcohol and two ounces of castor oil, and shake them together for fifteen minutes in a bottle. It will then be found that the alcohol has dissolved the oil, and the combination of the two makes a very excellent lotion for the hair. It can be perfumed with a few drops of the essential oils.



* Persons who write to us, expecting replies through this column, and those who may desire to make contributions to it of brief interesting facts, must always observe the strict rule, viz., to furnish their names, otherwise we cannot place confidence in their communications.

W. H., of N. Y.—We have no formulary for calculating the horse power of belts, and we believe it will be difficult to find any reliable rules for the purpose, because any belt can be made to transmit more or less power with the same velocity according to the pressure of the tightening pulley.

S. W., of Iowa.—You can keep up your window sash without the ordinary lines and pulleys by boring three or four holes in the sides of the sash, into which insert common bottle corks projecting about the sixteenth of an inch. These will press against the window frames along the usual groove, and by their elasticity support the sash at any required height.

G. C. T., of Pa.—The only way to clear a cellar of water which cannot be drained is to pump it out, then you should make a floor of hydraulic cement, and also coat the walls so as to prevent the future ingress of water. The anaesthetic process of printing would take up too much space to describe in this column.

G. B., of N. T.—When a magnetic needle is placed above or below a telegraph wire, it immediately deviates from its meridian, according to certain laws, and tends to place itself at right angles to the wire. This was discovered by Oersted in 1819. The person to whom you refer, who pretends he has just made such a discovery, is a Rip Van Winkle.

A. D. S., of Fulton.—The plan you propose for improving axle boxes is novel to us, and we believe a patent can be secured for it, if you can demonstrate its utility. We intended to write you, but know not the State in which you reside, as there are sixteen Fultons in the United States.

Pittston.—A letter from this place cannot be answered for want of the writer's name.

M. S., of N. Y.—You had better go to Auburn and examine Hutchinson's portable saw mill. We believe it is a good mill, and that it will answer your expectations.

INVENTOR.—The first patent granted by the United States of which there is any record was issued April 20, 1796, to Thos. Bidwell, for an improvement in forming yellow color. It was signed by Washington, Jefferson and Charles Lee. The next is dated Feb. 11, 1808, issued to Christian Jacob Hutter, for "a method of making brandy out of all kinds of grain or fruit equal in flavor, taste and color to the best imported French brandy."

R. C. S., of S. C.—You suggest that the difficulty with the Atlantic cable may be the pressure of the ocean through the pores of the gutta serena, thus injuring the insulation. We pointed out the possibility of such a contingency on page 29 of our last volume. At the depth of two miles the pressure on the square inch is over 4,000 pounds.

R. L., of S. C.—A tubular hydraulic telegraph is an old idea; it is inapplicable for ocean or land telegraphing.

B. S., of Md.—Your articles are ready to be set up, and we have prepared the engraving for the last one. In a week or two, at farthest, we shall commence their publication.

J. McB., of Mo.—Backlash is a technical term employed to denote counter resistance in machines, such as the back pressure of steam on a piston. It is not applied to the simple leakage of pistons, or loose parts of machinery, which merely cause increased friction or a loss of steam. It is not in any of the dictionaries known to us.

N. W. W., of N. Y.—The information received from those connected with the Atlantic Telegraph makes it appear that the currents sometimes pass very regularly for hours; then again they will cease entirely for days; and again they will be very feeble, and anon very strong. It is not easy to account for these phenomena; not as you suggest by a break of the conducting wires.

A. W., of Ill.—A "result" is not patentable except as "a new article of manufacture." A patented machine which embraces the means, devices or arrangements, for producing a certain result, does not cover the result itself; therefore a machine accomplishing a similar result in an entirely different manner is patentable, and does not infringe the rights of the other patentee. The second patent must not be a colorable evasion, but a substantially different invention.

SEA WATER.—Wherever there is shallow water, green will be produced by the underlying yellow sand, which, even in the absence of verdure on the shore or seaweeds beneath, always imparts a greenish tinge to the sea. The blue of the sky and the yellow of the sands meeting and intermingling in the water, form the green of the sea; the water acting as the medium in which the mixing or fusing of the colors takes place.

W. C., of N. Y.—Phosphorus is an animal production, and iodine a vegetable one. Phosphorus is the base of an acid found in bones, and the oxygen is detached by charcoal at a red heat. Iodine is procured from sea-weed. It is of a violet color, and evaporates very easily. It changes vegetable blues to yellow; is valuable as a medicine, and is much used for scrofulous complaints.

F. H., of Conn.—The Supreme Court of the United States, in the case of Hotchkiss vs. Greenwood cited in Howard's Reports, has clearly decided that the substitution of one material for another, even if the new material be better adapted to the purpose than the old, is not patentable.

D. B. C., of N. J.—You must be your own judge as to the propriety of paying \$15 for a superior engraving of your machine, and having it appear in a paper of 20,000 circulation, or to paying \$10 for an inferior one, and

have it published in a journal of only one or two thousand circulation. We do not insert engravings which have appeared in other American journals.

E. B., of Ohio.—If, according to your theory, the nucleus of a comet is a translucent sphere, and the tail produced simply by the sun's rays passing through the meridian, the same as through a lens, what is to become of the various comets which have no specific nucleus? Of course, if you were correct, there could be no such comets as that of Bruhna's.

H. J., of Ill.—Many persons injure their skin by applying cosmetics prepared by ignorant and dishonest persons, therefore you should beware. Spirits should not be applied to the skin as a practice. You can make a safe and excellent cosmetic where the skin requires a gentle stimulant as follows:—Take one ounce of scraped horseradish, and infuse for four hours in one pint of cold milk. Strain through muslin and settle, and apply with the hand or a piece of soft toweling.

Money received at the Scientific American Office on account of Patent Office business, for the week ending Saturday, October 9, 1858:—

- A. L. B., of Vt. \$30; M. & E., of Ill., \$25; W. T., of Ill., \$25; P. B., of N. Y., \$30; J. T., of N. Y., \$45; W. D., of N. J., \$20; I. R. & H. M. B., of N. Y., \$55; J. L. W., of Ohio, \$25; S. & B., of Ind., \$30; A. L., of N. Y., \$25; F. W., of —, \$15; J. O., of Pa., \$25; C. M., of N. Y., \$45; J. L., of Ind., \$11; J. C., of Mo., \$30; J. H. F., of Vt., \$30; J. H. T., of N. Y., \$100; W. H. R., of Fla., \$250; P. B., of N. Y., \$30; A. W. L., of Mass., \$25; T. H. M., of La., \$30; A. A., of N. Y., \$30; J. B. S., of Conn., \$30; I. H. G., of La., \$35; J. A. E., of N. Y., \$25; B. F. S., of Pa., \$25; F. R. N., of N. Y., \$250; W. H. B., of N. Y., \$30; McD. & M., of N. Y., \$33; W. & B., of Iowa, \$25; W. Van D., of Pa., \$60; S. M. B., of La., \$25; O. B. T., of Ohio, \$35; A. H. G., of N. Y., \$25; J. A. W., of Iowa, \$25; D. V., of Ohio, \$25; W. & T. S., of N. J., \$55.

Specifications and drawings belonging to parties with the following initials have been forwarded to the Patent Office during the week ending Saturday, October 9, 1858:—

- W. T., of Ill.; J. A. E., of N. Y.; M. B., of La.; J. L. W., of Ohio; S. M. B., of La.; C. M., of N. Y.; J. A. W., of Iowa; J. L., of Ind.; J. W. H., of L. I.; McD., & M., of N. Y.; D. V., of Ohio; J. E. S., of Me.; J. W. S., of Ohio; W. & T. S., of N. J.; A. H. G., of N. Y.; J. E., of N. Y.; M. & E., of Ill.; W. H., of Mass.; W. & B., of Iowa; W. Van D., of Pa. (2 cases); P. L., of N. Y.; O. B. T., of Ohio; A. W. L., of Mass.; J. J. H. B., of France; J. T., of N. Y.

Literary Notices.

BLACKWOOD'S MAGAZINE.—Published by L. Scott & Co., No. 54 Gold street.—The number for this month of this able magazine contains Bulwer's tale, "What will he do with it?" continued; and a very scientific essay on "Respiration and Suffocation." It contains several other articles of more than ordinary merit, among which is a new tale called "The Light on the Hearth."

THE DEMOCRATIC AGE.—Edited by C. Edwards Lester, No. 40 Park Row, New York.—This new magazine and review contains many excellent articles, and the first number promises well. It is published monthly, each number containing about 80 pages, and the terms are only \$3 per annum.

THE AMERICAN JOURNAL OF PHOTOGRAPHY AND THE ALLIED ARTS.—The title of this publication fully indicates its nature and character. It is edited and published semi-monthly by Charles A. Seely, practical chemist and manufacturer of materials employed in the photographic art. It contains all the useful information to be had in reference to the progress of this beautiful art. Terms, \$1 50 per annum. Office, No. 424 Broadway.

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RECEIPTS.—When money is paid at the office for subscriptions, a receipt for it will always be given; but when subscribers remit their money by mail, they may consider the arrival of the first paper a bona fide acknowledgment of the receipt of their funds. The Post Office law does not allow publishers to enclose receipts in the paper.

PATENT CLAIMS.—Persons desiring the claim of any invention which has been patented within fifteen years, can obtain a copy by addressing a letter to this office, stating the name of the patentee, and date of patent when known, and enclosing \$1 as fee for copying.

BINDING.—We would suggest to those who desire to have their volumes bound, that they had better send their numbers to this office, and have them executed in a uniform style with their previous volumes. Price of binding 75 cents.

IMPORTANT TO INVENTORS.

AMERICAN AND FOREIGN PATENT SOLICITORS.—Messrs. MUNN & CO., Proprietors of the Scientific American, continue to procure patents for inventors in the United States and all foreign countries on the most liberal terms. Our experience is of thirteen years' standing, and our facilities are unequalled by any other agency in the world. The long experience we have had in preparing specifications and drawings has rendered us perfectly conversant with the mode of doing business at the United States Patent Office, and with most of the inventions which have been patented. Information concerning the patentability of inventions is freely given, without charge, on sending a model or drawing and description to this office. Consultation may be had with the firm, between nine and four o'clock, daily, at their principal office, 128 Fulton street, New York. We established, over a year ago, a Branch Office in the City of Washington, on the corner of F and Seventh streets, opposite the United States Patent Office. This office is under the general superintendence of one of the firm, and is in daily communication with the Principal Office in New York, and personal attention will be given at the Patent Office to all such cases as may require it. Inventors and others who may visit Washington, having business at the Patent Office, are cordially invited to call at our office.

Inventors will do well to bear in mind that the English law does not limit the issue of patents to inventors. Any one can take out a patent there. We are very extensively engaged in the preparation and securing of patents in the various European countries. For the transaction of this business we have offices at Nos. 66 Chancery Lane, London; 29 Boulevard St. Martin, Paris; and 26 Rue des Eperonniers, Brussels. We think we may safely say that three-fourths of all the European patents secured to American citizens are procured through our Agency. Circulars of information concerning the proper course to be pursued in obtaining patents through our Agency,

the requirements of the Patent Office, &c., may be had gratis upon application at the principal office or either of the branches.

The annexed letter from the late Commissioner of Patents we commend to the perusal of all persons interested in obtaining patents:— Messrs. MUNN & CO.—I take pleasure in stating that while I held the office of Commissioner of Patents, MORE THAN ONE-FOURTH OF ALL THE BUSINESS OF THE OFFICE came through your hands. I have no doubt that the public confidence thus indicated has been fully deserved, as I have always observed, in all your intercourse with the Office, a marked degree of promptness, skill, and fidelity to the interests of your employers. Yours, very truly, CHAS. MASON. Communications and remittances should be addressed to MUNN & COMPANY, No. 138 Fulton street, New York.

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MACKINTOSH & WADSWORTH'S PATENT Variable Governor Cut-off Valve, equally adapted to the common slide valve or puppet valve engines, as to the oscillating; cutting off the steam at any point, from the commencement to three-fourths of the stroke, as the varying pressure of the steam in the boiler, or the varying amount of work to be done, requires. Shop, county and State rights for sale. For illustration see Soc. Am., Vol. XIII, No. 51. For full particulars address CRIDGE, WADSWORTH & CO., Pittsburgh, Pa. 3 tf

WOODBURY'S IMPROVED WOODWORTH Planing, Tonguing and Grooving Machines, are warranted to be vastly superior to any other machines in this country. When exhibited, they have always received the highest premium. Two gold medals have been awarded. Six patents have been granted to secure the improvements on these machines. All sizes constantly for sale, by JAMES A. WOODBURY, 69 Sudbury street, Boston. 1 8

Science and Art.

Peculiarities of Color and Temperature of the Ocean.

It is a commonly observed fact that the usual color of the ocean is a bluish green, of a darker tint at a distance from land, and clearer towards the shores. According to Dr. Scoresby, the hue of the Greenland sea varies from ultramarine blue to olive green, and from the purest transparency to great opacity. The surface of the Mediterranean, in its upper part, is said to have, at times, a purple tint. In the gulf of Guinea the sea sometimes appears white, about the Maldive islands black, and near California it has a reddish appearance. Various causes must, of course, co-operate to produce this diversity of tint. The prevailing blue color is generally ascribed to the greater-refrangibility of the blue rays of light, which, by reason of that property, pass in greatest abundance through the water. The other colors are ascribed to the existence of vast numbers of minute animalculæ—to marine vegetables at or near the surface—to the color of the soil—the infusion of earthy substances—and very frequently the tint is modified by the aspect of the sky. The phosphorescent, or slimy appearance of the ocean, which is a common phenomenon, is also ascribed to animalculæ and to semi-putrescent matter diffused through the water.

The temperature of the ocean also exhibits some peculiar and interesting phenomena. Within the tropics the mean temperature at the surface is about 80° Fah., and generally ranges between 77° and 84°. At great depths the temperature is probably nearly the same under every latitude. In the torrid zone it is found to diminish with the depth, while in the polar seas it increases with the depth; and about the latitude of 70° it is nearly constant at all depths. But the small number of observations which have yet been made on this subject do not indicate any uniform law, according to which the variations of temperature at different depths is regulated.—*Ex.*

Improved Carriage Shafts.

This invention is designed to enable one or two horses to be attached to a vehicle, the shafts being capable of being made into a pole by simply closing them.

Our illustration and description will fully elucidate the invention, Fig. 1 being the arrangement as shafts, and Fig. 2 as a pole. Fig. 3 is the shaft and pole attachment detached.

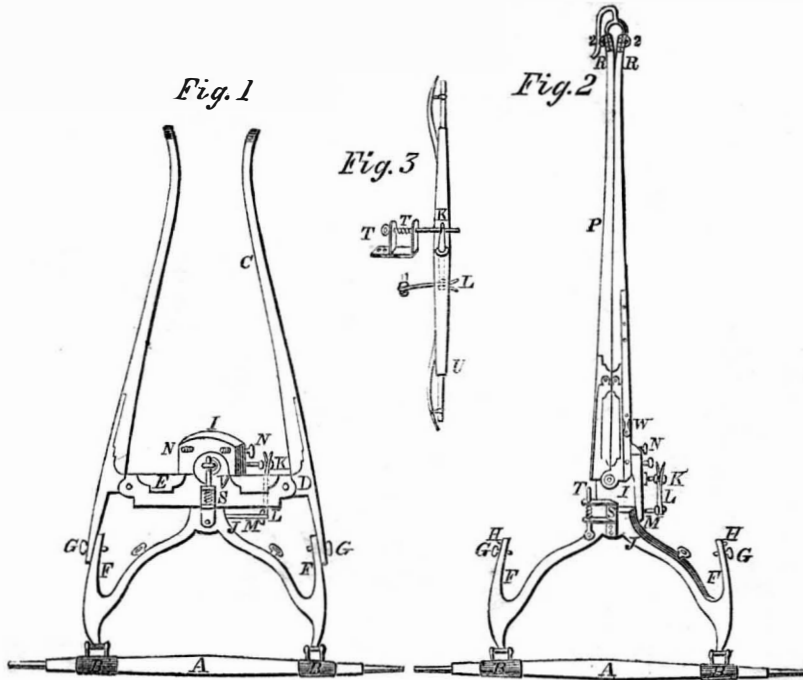
A represents the carriage axle, and B the clips, to which the shafts are usually attached. C are the shafts, having an iron knuckle joint, D, at each side, by which they are connected to the cross piece, E. The front end of joint D, is permanently screwed to the shaft, and the other end extends back to the pole bar point, F, where it is attached by a thumb screw, G, and pin, H. I represents the iron plate, which is fastened to the front end of pole bar, J. Through the center of the plate, I, is a main bolt, K, which is held down at its lower end by a forked spring, L, the spring being attached at its back end to the pin, M. N N are two thumb screws near the front of the plate, for the purpose of being screwed up against the shafts, C, to tighten them when they are closed together and form the pole, P. At the front end of the pole is a hook, having two eyes, 2, and its points, R, fitting into the ends (in a socket) of pole or tongue, with a shoulder on each, so that the horses can be hitched by the pole straps to the eyes, 2, to hold back the carriage, but in case of the horses becoming unruly and running off, they are detached from the swiveltree behind, and in moving forward the hook slips out, and frees the horses from the pole.

S represents a plate with two uprights, through which a disconnecting pin with a

spiral spring, T, passes. This pin, T, is intended to hold the swiveltree, U, in its place. The bolt, K, is pressed upwards by the operator through the center of the swiveltree, U, and then the pin, T, is forced forward by its spiral spring through an eye in the upper

end of bolt, K. The swivel or doubletree, U, is thus fastened, but in the event of the horse or horses running off, the driver pulls a strap which is attached to the head of the pin, T, and as the pin, T, is drawn back, the bolt, K, falls down, and the swiveltree is instantly

HOFFMEIER'S SHAFTS FOR VEHICLES.



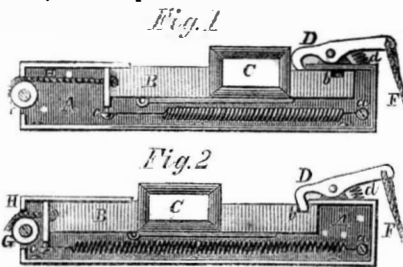
disengaged, and the horse is at once loosed from the carriage. The cross-piece, E, has a joint, V, in its center, and the bolt, K, passes through the joint, V. When the ends of the shaft at its connection, F, with the pole bar, J, are unscrewed by the screw, G, the shafts will operate and close at the joint, V, and

the ends of shaft will hook on to the thumb screws, W.

They are the invention of A. K. Hoffmeier, of Lancaster, Pa., and were patented by him September, 7, 1858. He will be happy to furnish any further information upon being addressed as above.

Wells' Belt-Shipper.

The ordinary belt-shippers are by no means secure, and the belt which drives machinery from a main pulley has often become changed from a "fast" to a "loose" pulley by accident, thus causing serious consequences to the piece of wood or metal being operated upon, and the machine tool then in use. To prevent any future accidents of this kind, Morris Wells, of Brooklyn, E. D., in this State, has invented the belt-shipper which is the subject of our engraving, and at the same time to furnish a cheap and sure belt-shipper, that can be depended upon in all cases. It is simple, small, and compact.



Our illustrations show it with the front plate off, that its working parts may be seen, Fig. 1 being as it would keep the belt on one pulley, and Fig. 2 as it would keep the belt on the other.

It consists of a cast iron box, A, in which the bolt, B, is free to slide back and forth on a projecting piece cast with A, and seen under it. The case, A, can be secured by screws to any beam or piece of metal or wood over which the belt may pass, at any angle, or in any position to suit the direction of the belt. In B is a square slot, C, rounded on its interior surface, so as not to cut the belt which passes through it.

The operation is simple. Suppose the belt to be on the loose pulley, and the shipper in the position seen in Fig. 1, the bolt, B, would be held securely in that position by the spring which is attached to it at c, and to the box at a. When it was desired to run the machine, the cord, H, which may be of any length, and conveyed any distance, and which passes over the small pulley, G, in the box, must be

pulled, and the spring would be distended, while the bolt would assume the position shown in Fig. 2, and there it would be held by the end of catch, D, being forced into slot b, of the bolt by the spring, d; this brings the belt on to the fast pulley. Should it then be convenient to stop the machine, the cord, F, must be pulled. This depresses one end of the catch, and elevates the other, releasing the bolt, which the spring draws back, and with it the belt, on to the loose pulley of the machine.

This is an addition to the machine shop that has long been wanted, and we are happy in being able to recommend this invention. It was patented February 2, 1858, and the inventor will furnish any further information upon being addressed as above.

Curing Hams.

As the time is at hand for preparing these useful stores of rich and savory food, a few words will not be out of place in regard to them. The legs of hogs, short in the hock, are the best for hams, and should be chosen in preference to lanky legs. They may be salted by immersion in a clean pickle, containing a little sugar and saltpeter dissolved, or they may be salted by rubbing ground solar evaporated salt over them, turning them every day, and giving them a good rubbing. A little sugar and ground black pepper added to the salt will much improve the flavor of the meat. It requires about a month to salt hams by the wet process, and three weeks by the dry system. At the end of this period, they should be hung up for a few days to drip, and then they are ready for smoking. Much depends on the kind of material used for smoking them, so as to secure a sweet flavor. Whatever fuel is used for this purpose, one condition should never be overlooked; it should be perfectly dry, or else it will be liable to impart a bitter taste to the meat. Dry corn cobs, and some dry sweet hay are superior to all other agents that we have seen employed for smoking beef and hams.

Mutton hams may be prepared in the same manner as those of pork, and they are ex-

ceedingly palatable when the meat is good, and care exercised to smoke them slowly.

First Employment of Coal as Fuel.

As an evidence of the vast difficulty experienced by introducers of new articles, from the prejudices of a community alone, we may mention a fact in relation to the employment of the useful material of coal as a fuel. When coal was first introduced into England as a fuel, the prejudice against it was so strong that the Commons petitioned the Crown to prohibit the "noxious" fuel. A royal proclamation having failed to abate the nuisance, a commission was issued to ascertain who burned coal within the city of London and its neighborhood, and to punish them by fine for the first offence, and by demolition of their furnaces if they persisted in transgressing. A law was finally passed making it a capital offence to burn coal in the city, and only permitting it to be used in the forges in the vicinity. It is stated that among the records in the town of London, a document was once found purporting that in the time of Edward I. a man had been tried, convicted and executed for the crime of burning coal in London! It took three centuries to entirely efface this prejudice.

SPONTANEOUS COMBUSTION.—A material much used for flooring and roofing in Europe, and called "asphalted felt," has on various occasions been found to be on fire. All such materials as are composed of organic fibers mixed with hydro-carbons should be carefully used, as they are all liable to become ignited from chemical decomposition.



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